

Name :

Date :

# Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

- i) Which element is the atom shown above? \_\_\_\_\_
- ii) What is the chemical symbol of the element? \_\_\_\_\_
- iii) What is the mass of the atom shown above? \_\_\_\_\_
- iv) How many valence electrons are shown in the atom above? \_\_\_\_\_



② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? \_\_\_\_\_

③ The nucleus was discovered by which scientist:

- a) Thomson                      b) Dalton                      c) Rutherford                      d) Democritus

④ What is the mass of Carbon-14: \_\_\_\_\_

⑤ What is the number of protons in Carbon-14: \_\_\_\_\_

⑥ What is the number of neutrons in Carbon-14: \_\_\_\_\_

⑦ What is the number of protons in the radioactive Uranium-238: \_\_\_\_\_

⑧ What is the number of neutrons in Sulfur-33: \_\_\_\_\_

⑨ Neutral atoms of the same element can differ in

- a) Mass number                      b) Atomic number                      c) Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

- a) Avogadro                      b) Bohr                      c) Dalton

# Atomic Structure Chapter Worksheet

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### **Table of Contents Atomic Structure Chapter Worksheet**

1. Understanding the eBook Atomic Structure Chapter Worksheet
  - The Rise of Digital Reading Atomic Structure Chapter Worksheet
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Chapter Worksheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Structure Chapter Worksheet
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Chapter Worksheet
  - Personalized Recommendations
  - Atomic Structure Chapter Worksheet User Reviews and Ratings
  - Atomic Structure Chapter Worksheet and Bestseller Lists

5. Accessing Atomic Structure Chapter Worksheet Free and Paid eBooks
  - Atomic Structure Chapter Worksheet Public Domain eBooks
  - Atomic Structure Chapter Worksheet eBook Subscription Services
  - Atomic Structure Chapter Worksheet Budget-Friendly Options
6. Navigating Atomic Structure Chapter Worksheet eBook Formats
  - ePub, PDF, MOBI, and More
  - Atomic Structure Chapter Worksheet Compatibility with Devices
  - Atomic Structure Chapter Worksheet Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Atomic Structure Chapter Worksheet
  - Highlighting and Note-Taking Atomic Structure Chapter Worksheet
  - Interactive Elements Atomic Structure Chapter Worksheet
8. Staying Engaged with Atomic Structure Chapter Worksheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Atomic Structure Chapter Worksheet
9. Balancing eBooks and Physical Books Atomic Structure Chapter Worksheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Atomic Structure Chapter Worksheet
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Chapter Worksheet
  - Setting Reading Goals Atomic Structure Chapter Worksheet
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure Chapter Worksheet
  - Fact-Checking eBook Content of Atomic Structure Chapter Worksheet
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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