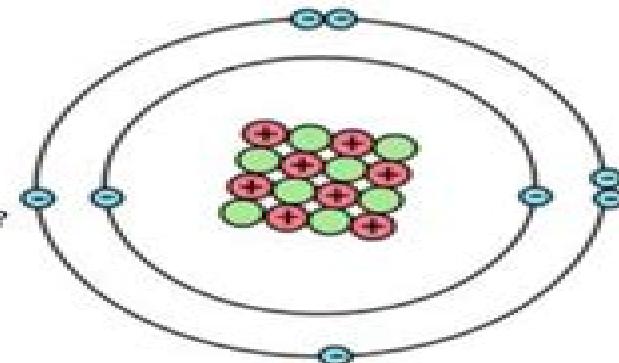


# Atomic Theory – Unit Exam

For all of the following multiple-choice questions, choose only one correct answer:

1. Which description below accurately describes the current model of the atom?
  - a) Electrons are distributed around the nucleus and have almost no mass compared to protons
  - b) Protons, electrons and neutrons are evenly distributed throughout all parts of the atom
  - c) The nucleus consists of protons, electrons and neutrons
  - d) The nucleus is made of electrons and protons
2. Which of the following is **true** about subatomic particles?
  - a) Electrons are negatively charged and are the heaviest subatomic particle
  - b) The mass of a proton nearly equals the mass of a neutron
  - c) Protons are positively charged and the lightest subatomic particle
  - d) Neutrons are negatively charged and are the heaviest subatomic particle
3. What would be the **mass** of the atom pictured to the right?
  - a) 7 amu
  - b) 8 amu
  - c) 16 amu
  - d) 24 amu
4. How many **valence electrons** are shown in the atom on the right?
  - a) 2 valence electrons
  - b) 4 valence electrons
  - c) 6 valence electrons
  - d) 8 valence electrons
5. Why are atoms electrically **neutral**?
  - a) The number of neutrons are equal to the number of protons and electrons.
  - b) The numbers of neutrons and electrons are equal and cancel each other out.
  - c) The numbers of protons and electrons are equal and cancel each other out.
  - d) The numbers of protons and neutrons are equal and cancel each other out.
6. Based on the symbol to the right, what can be concluded about the element **fluorine**?
  - a) There are 19 atoms in a molecule of fluorine
  - b) There are 9 neutrons outside of the fluorine atom's nucleus
  - c) There are 9 protons in the nucleus of a fluorine atom
  - d) There are 9 valence electrons in the fluorine atom.



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Fluorine
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# Atomic Theory Unit Regents Review Answers

**JG Myers**

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## **Table of Contents Atomic Theory Unit Regents Review Answers**

1. Understanding the eBook Atomic Theory Unit Regents Review Answers
  - The Rise of Digital Reading Atomic Theory Unit Regents Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Theory Unit Regents Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Theory Unit Regents Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Theory Unit Regents Review Answers

- Personalized Recommendations
- Atomic Theory Unit Regents Review Answers User Reviews and Ratings
- Atomic Theory Unit Regents Review Answers and Bestseller Lists

5. Accessing Atomic Theory Unit Regents Review Answers Free and Paid eBooks

- Atomic Theory Unit Regents Review Answers Public Domain eBooks
- Atomic Theory Unit Regents Review Answers eBook Subscription Services
- Atomic Theory Unit Regents Review Answers Budget-Friendly Options

6. Navigating Atomic Theory Unit Regents Review Answers eBook Formats

- ePUB, PDF, MOBI, and More
- Atomic Theory Unit Regents Review Answers Compatibility with Devices
- Atomic Theory Unit Regents Review Answers Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Atomic Theory Unit Regents Review Answers
- Highlighting and Note-Taking Atomic Theory Unit Regents Review Answers
- Interactive Elements Atomic Theory Unit Regents Review Answers

8. Staying Engaged with Atomic Theory Unit Regents Review Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Atomic Theory Unit Regents Review Answers

9. Balancing eBooks and Physical Books Atomic Theory Unit Regents Review Answers

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Atomic Theory Unit Regents Review Answers

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Atomic Theory Unit Regents Review Answers

- Setting Reading Goals Atomic Theory Unit Regents Review Answers
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Atomic Theory Unit Regents Review Answers

- Fact-Checking eBook Content of Atomic Theory Unit Regents Review Answers
- Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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