

Atomic Radiation and Half-Life

100 pts

multiple choice

Read the following scenario and the following questions. You must answer all questions to receive full credit for this section.

1. Carbon-14 dating is used to determine the age of what type of object?
 - a. Coal
 - b. Coal ash
 - c. Coal smoke
 - d. Coal dust
2. The half-life of a radioactive isotope is the time it takes for the isotope to decay to half of its original amount.
 - a. True
 - b. False
3. Complete the following table.

Isotope	Initial Amount	Half-life	Half-life of Isotope	Remaining Amount	Half-life of Isotope
U-238	100	4.5 billion years	4.5 billion years	100	4.5 billion years
U-238	100	4.5 billion years	4.5 billion years	50	4.5 billion years
U-238	100	4.5 billion years	4.5 billion years	25	4.5 billion years
U-238	100	4.5 billion years	4.5 billion years	12.5	4.5 billion years

4. Refer to the table above. After the isotope has decayed for 10.5 billion years, what is the remaining amount?
 - a. 100
 - b. 50
 - c. 25
 - d. 12.5
5. Refer to the table above. After the isotope has decayed for 10.5 billion years, what is the remaining amount?
 - a. 100
 - b. 50
 - c. 25
 - d. 12.5

ANSWER: Multiple Choice: 100% (100% correct)

Multiple Choice: There are three different ways to calculate how long it takes for a radioactive isotope to decay to half of its original amount. The following are three different ways to calculate the half-life of a radioactive isotope. Write each method on a separate line. Then, write a brief description of each method. If the first method is right, the others are either wrong or have imperfections. The descriptions below will help you write the descriptions of the other methods.

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