

Atomic Radiation and Half-Life

2000 pts

2000-1-1-10

Read the entire laboratory investigation and the relevant pages of your textbook, discussing it with your lab group.

1. Explain half-life, activity, radiations, isotopes, and radioactive decay.
2. The first part of the investigation is being done in class. What is the goal?
3. Complete the following table:

TABLE 1. RADIOISOTOPES					
Isotope	Atomic Number	Atomic Mass	Number of Protons	Number of Neutrons	Number of Electrons
$^{131}_{53}\text{I}$					
$^{137}_{55}\text{Cs}$					
$^{132}_{54}\text{Xe}$					
$^{131}_{52}\text{Te}$					
$^{131}_{54}\text{Xe}$					

4. What is the most accurate the recording of the counts?
5. What is the difference between gamma radiation and X-ray radiation?

APPENDIX B: RADIATION AND HALF-LIFE LAB

Radioisotopes undergo the spontaneous change in α , β , γ , and electron capture decays as well as half-life. From the studying of decays, there has been a great interest in understanding isotopes and their natural state, with half-life being a key in The Radioactive Clock, the Atom, and other popular books. Furthermore, the purpose of this laboratory activity is to explore the different types of radiation, half-life, and the associated nature of radioactivity.

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