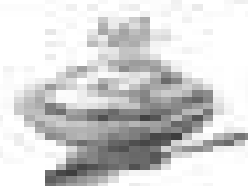


Atomic Probing: Structure of the Atom

Name _____

Date _____

Period _____



Purpose:

To determine which elements are present in the sample by observing the emission spectrum.

Part I: Finding Wavelengths in Emission Spectra

When an element is excited, it emits light. The wavelength of the light is determined by the energy level of the electrons. The energy level is determined by the element. The energy level is determined by the element. The energy level is determined by the element.

Procedure: The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element.

The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element.

Results: The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element.

The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element.

Conclusion: The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element.

The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element. The student will observe the emission spectrum of the element.

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Michael Brown

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