

# Pre-lab Questions for Experiment #5

## Rates of Chemical Reactions I: A Clock Reaction

1) Complete and balance the following reactions:



2) The rate law for the reaction



is  $\text{rate} = k[\text{H}_2][\text{NO}]^2$ . Which of the following mechanisms can be ruled out on the basis of the observed rate expression?

Mechanism I



Mechanism II



Mechanism III



3) Consider the reaction



From the following data obtained at a certain temperature, determine the order of the reaction and calculate the rate constant:

[A] (M)	[B] (M)	Rate (M/s)
1.50	1.50	$3.20 \times 10^{-3}$
1.50	2.50	$3.20 \times 10^{-3}$
3.00	1.50	$6.40 \times 10^{-3}$

4) The following gas-phase reaction was studied at 290°C by observing the change in pressure as a function of time in a constant-volume vessel:



Determine the order of the reaction and the rate constant based on the following data:

Time (s)	Partial Pressure of $\text{ClCO}_2\text{CCl}_3$ (mmHg)
0	15.76
181	12.64
513	8.73
1164	4.44

## Answer Of Pre Lab Rates Of Reaction

**Wendy Gloffke,Doris Kimbrough**

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**Chemistry** Lemay,Prentice-Hall Staff,1996    **Integrated Approach to Coordination Chemistry** Rosemary A. Marusak,Kate Doan,Scott D. Cummings,2007-03-07 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials    **Chemistry** McGraw-Hill Staff,2001-07    **Pre-lab Exercises for Experimental Organic Chemistry** Royston M. Roberts,John C. Gilbert,Stephen F. Martin,1994    **Pre-Lab Exercises for Modern Experimental Organic Chemistry** Royston M. Roberts,1985    **Introductory Chemistry** Wendy Gloffke,Doris Kimbrough,2002 This lab manual helps students develop data acquisition organization and analysis skills while teaching basic techniques Students construct their own data tables answer conceptual questions and make predictions before performing experiments They also have the opportunity to visualize and describe molecular level activity and explain the results    **Laboratory Experiments for General Chemistry** Harold R. Hunt,Toby F. Block,George M. McKelvy,1998    **Experimental Organic Chemistry** Daniel R. Palleros,2000-02-04 This cutting edge lab manual takes a multiscale approach presenting both micro semi micro and macroscale techniques The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion    **Chemical Investigations** Nancy Konigsberg Kerner,1986    **Experimental Organic Chemistry** John C. Gilbert,Stephen F. Martin,2006 This proven and well tested laboratory manual for organic chemistry students contains procedures for both miniscale also known as small scale and microscale users This lab manual gives students all the necessary background to enter the laboratory with the knowledge to perform the experiments with confidence For the microscale labs experiments were chosen to provide tangible quantities of material which can then be analyzed Chapters 1 2 introduce students to the equipment record keeping and safety of the laboratory Chapters 3 6 and 8 are designed to introduce students to laboratory techniques needed to perform all experiments In Chapters 7 and 9 through 20 students are required to use the techniques to synthesize compounds and analyze their properties In Chapter 21 students are introduced to multi step syntheses of organic compounds a practice well known in chemical industry In Chapter 23 students are asked to solve structures of unknown compounds The new chapter 24 introduces a meaningful experiment into the textbook that reflects the increasing emphasis

on bioorganic chemistry in the sophomore level organic lecture course This experiment not only gives students the opportunity to accomplish a mechanistically interesting and synthetically important coupling of two amino acids to produce a dipeptide but also provides valuable experience regarding the role of protecting groups in effecting synthetic transformations with multiple functionalized molecules

**Methods in Biotechnology** Seung-Beom Hong,M. Bazlur

Rashid,Lory Z. Santiago-Vázquez,2016-08-01 As rapid advances in biotechnology occur there is a need for a pedagogical tool to aid current students and laboratory professionals in biotechnological methods Methods in Biotechnology is an invaluable resource for those students and professionals Methods in Biotechnology engages the reader by implementing an active learning approach provided advanced study questions as well as pre and post lab questions for each lab protocol These self directed study sections encourage the reader to not just perform experiments but to engage with the material on a higher level utilizing critical thinking and troubleshooting skills This text is broken into three sections based on level Methods in Biotechnology Advanced Methods in Biotechnology I and Advanced Methods in Biotechnology II Each section contains 14 22 lab exercises with instructor notes in appendices as well as an answer guide as a part of the book companion site This text will be an excellent resource for both students and laboratory professionals in the biotechnology field

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web i remembered the features much better than i though i would funny how what one learns in childhood gets burned in seeing valles alpis top left of left hand photo was especially

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web w what ang well we were hiding in the trees weren t we j that s right we didn t see nothing p we don t know nothing about it do us ray poor old quack quack into my

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web peter a cry baby john one puncll and him ll give in one punch that s all the barn peter flapping his arms quack quack donald duck left alone sits huddled up against the

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