

## SECOND TEST DECEMBER-2023

CLASS : I PUC  
TIME : 1 Hr. 30 Mins.

SUBJECT : CHEMISTRY (34)

Max. Marks : 35

## PART-A

I Select the correct option from the given choices :  $6 \times 1 = 6$

- Relation between  $K_p$  and  $K_c$  is given by
  - $K_p = K_c (RT)^{\Delta n}$
  - $K_p = K_c (RT)^{\Delta n}$
  - $K_p = K_c / (RT)^{\Delta n}$
  - $K_p = (RT)^{\Delta n} / K_c$
- At 298 K,  $K_a \times K_b$  is
  - $10^{-2}$
  - $10^{-10}$
  - $10^{-10}$
  - $10^{-7}$
- The process of addition of electrons is called
  - Reduction
  - Oxidation
  - Displacement reaction
  - Decomposition reaction
- The oxidation number of an atom in the elemental state is
  - +1
  - 1
  - zero
  - $-\frac{1}{2}$
- Soda lime is a mixture of
  - $\text{NaOH} + \text{KOH}$
  - $\text{NaOH} + \text{CaO}$
  - $\text{NaOH} + \text{Ca(OH)}_2$
  - All of these
- In benzene each carbon atom undergoes
  - $sp^2$  hybridization
  - $sp^2$  hybridization
  - $sp$  hybridization
  - None of these

II Fill in the blanks by choosing appropriate word from those given in brackets:  $3 \times 1 = 3$

[ $\text{C}_6\text{H}_{2n-2}$ ,  $\text{C}_n\text{H}_{2n}$ , Zero,  $\text{HSO}_4^-$ ]

- Conjugate base of  $\text{H}_2\text{SO}_4$  is \_\_\_\_\_.
- The standard electrode potential of standard hydrogen electrode is \_\_\_\_\_.
- The general formula of alkyne is \_\_\_\_\_.

## PART-B

III Answer any TWO of the following questions. Each question carries **TWO** marks:  $2 \times 2 = 4$

- What is heterogeneous equilibrium ? Give an example.
- What is conjugate acid-base pair ? Illustrate with an example.
- Explain the preparation of alkenes by dehydrohalogenation of alkyl halides.
- Write the Newman's projection of ethane.

## PART-C

IV Answer any TWO of the following questions. Each question carries **THREE** marks:  $2 \times 3 = 6$

- Balance the following redox reaction by oxidation number method in acidic medium  
 $\text{Cr}_2\text{O}_7^{2-} + \text{SO}_3^{2-} \longrightarrow \text{Cr}^{3+} + \text{SO}_4^{2-} + \text{H}_2\text{O} -$
- Define oxidising agent. Give one example each for oxidising and reducing agents. (P.T.O)

## 2nd Puc Cbse Question Paper

**Prof. S. K. Khanna, Dr. N. K. Verma, Dr.  
B. Kapila**

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observed that knowledge alone is not sufficient to achieve high marks One needs to learn the skills to translate knowledge into performance on examination day For this purpose we feel immense pleasure in introducing this 10 Sample Question Papers for CBSE Board Term 2 examination These papers have been designed keeping in mind the latest pattern of examination These question papers provide opportunity to student to practice in examination like environment by practicing 10 times before the actual examinations Practicing these paper improves student exam taking skills and scoring ability It also builds student s examination temperament Key features of the books are All papers are based on latest syllabus pattern marking scheme of CBSE Term 2 Solutions to all papers are given Latest CBSE Term 2 Sample Question Paper Based on Latest Competency Based Education CBE Students are advised to take all papers in examination like environment with full sincerity in prescribed time limit so that they get fully tuned for the examination environment and ensure very good performance in exams

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