

17. Arrange the following elements in the order of their decreasing metallic character

Na, Si, Cl, Mg, Al

- (a)  $\text{Cl} > \text{Si} > \text{Al} > \text{Mg} > \text{Na}$
- (b)  $\text{Na} > \text{Mg} > \text{Al} > \text{Si} > \text{Cl}$
- (c)  $\text{Na} > \text{Al} > \text{Mg} > \text{Cl} > \text{Si}$
- (d)  $\text{Al} > \text{Na} > \text{Si} > \text{Ca} > \text{Mg}$

18. Arrange the following elements in the order of their increasing non-metallic character

Li, O, C, Be, F

- (a)  $\text{F} < \text{O} < \text{C} < \text{Be} < \text{Li}$
- (b)  $\text{Li} < \text{Be} < \text{C} < \text{O} < \text{F}$
- (c)  $\text{F} < \text{O} < \text{C} < \text{Be} < \text{Li}$
- (d)  $\text{F} < \text{O} < \text{Be} < \text{C} < \text{Li}$

19. What type of oxide would Eka-aluminium form?

- (a)  $\text{EO}_3$
- (b)  $\text{E}_3\text{O}_2$
- (c)  $\text{E}_2\text{O}_3$
- (d)  $\text{EO}$

20. Three elements B, Si and Ge are

- (a) metals
- (b) non-metals
- (c) metalloids
- (d) metal, non-metal and metalloid respectively

21. Which of the following elements will form an acidic oxide?

- (a) An element with atomic number 7
- (b) An element with atomic number 3
- (c) An element with atomic number 12
- (d) An element with atomic number 19

22. The element with atomic number 14 is hard and forms acidic oxide and a covalent halide. To which of the following categories does the element belong?

- (a) Metal
- (b) Metalloid
- (c) Non-metal
- (d) Left-hand side element

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