

IB SL and HL Topic 1 to 5 Chemistry MCQ
40 Marks

1. What amount of oxygen, O_2 , (in moles) contains 1.8×10^{22} molecules?

A. 0.0030
B. 0.030
C. 0.30
D. 3.0

(Total 1 mark)

2. Which compound has the empirical formula with the greatest mass?

A. C_2H_6
B. C_6H_{10}
C. C_5H_{10}
D. C_8H_8

(Total 1 mark)



When the equation above is balanced, what is the coefficient for oxygen?

A. 2
B. 3
C. 4
D. 5

(Total 1 mark)

4. 3.0 dm^3 of sulfur dioxide is reacted with 2.0 dm^3 of oxygen according to the equation below.



What volume of sulfur trioxide (in dm^3) is formed? (Assume the reaction goes to completion and all gases are measured at the same temperature and pressure.)

A. 5.0
B. 4.0
C. 3.0
D. 2.0

(Total 1 mark)

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