

16. Increasing the temperature generally.....
- decreases the solubility of a solid in a liquid but increases the solubility of a gas in a liquid.
  - increases the solubility of a solid in a liquid but decreases the solubility of a gas in a liquid.
  - increases the solubility of both a solid and a gas in a liquid.
  - decreases the solubility of both a solid and a gas in a liquid.
17. A white precipitate was formed when  $\text{BaCl}_{2(aq)}$  was added to an aqueous solution of a salt X. The precipitate dissolved in dilute  $\text{HCl}$  with rapid effervescence. Salt X is likely to contain
- $\text{SO}_4^{2-}$  ions.
  - $\text{SO}_3^{2-}$  ions.
  - $\text{NO}_3^-$  ions.
  - $\text{CO}_3^{2-}$  ions.
18. Before a reaction could take place, there should be
- ionization of reactant particles.
  - breakage of bonds of products.
  - breakage of bonds of reactants.
  - ionization of product particles.
19. Consider the following reaction equation:  
 $\text{CaO} + \text{SiO}_2 \rightarrow \text{CaSiO}_3$   
 Silicon(IV) oxide is acting as
- a basic oxide.
  - a reducing agent.
  - an acidic oxide.
  - an oxidizing agent.
20. Which of the following acids would form normal salt only?
- Tetraoxosulphate(VI) acid
  - Trioxonitrate(V) acid
  - Tetraoxophosphate(V) acid
  - Trioxosulphate(IV) acid
21. What is the partial pressure of oxygen at s.t.p. in a gaseous mixture containing  $100 \text{ cm}^3$  of oxygen and  $900 \text{ cm}^3$  of nitrogen gas?
- 1.0 atm
  - 0.1 atm
  - 9.0 atm
  - 0.9 atm
22. Graphite is used as a dry lubricant due to the presence of
- mobile electrons.
  - free electrons.
  - octahedral structures.
  - layered structures.
23. Which of the following gases is alkaline?
- $\text{NO}_2$
  - $\text{CO}_2$
  - $\text{NH}_3$
  - $\text{N}_2\text{O}$
24. Which of the following statements about an equilibrium system is correct?
- Forward and backward reactions occur at the same rate.
  - The concentrations of reactants must equal that of the products.
  - The concentrations of reactants and products can be changed by adding a catalyst.
  - The concentrations of reactants and products are not affected by a change in temperature.

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*School Science Practical Work in Africa* Umesh Ramnarain, 2020-06-01 School Science Practical Work in Africa presents the scope of research and practice of science practical work in African schools. It brings together prominent science educators and researchers from Africa to share their experience and findings on pedagogical innovations and research-informed practices on school science practical work. The book highlights trends and patterns in the enactment and role of practical work across African countries. Practical work is regarded as intrinsic to science teaching and learning, and the form of practical work that is strongly advocated is inquiry-based learning, which signals a definite paradigm shift from the

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