

Nuclear Radiation Worksheet

1. a. What determines whether an atom is stable or unstable?
Its overall size and a balance between protons and neutrons.

b. How does an unstable atom gain stability?
It undergoes radioactive decay.

c. What part of an atom undergoes change during radioactive decay?
The nucleus.

2. Which of the human senses can detect radiation?

Our eyes can detect colors, but we have no sense to detect alpha, beta, or gamma radiation.

3. Explain the difference between an isotope and a radioactive isotope.

An isotope has a different set of protons and neutrons. Radioactive isotope is large unstable, and has a short life.

4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	Four protons	Positive (+)	Stopped by skin or paper
Beta	One Neutron (isotrons)	Negative (-)	Stopped by sheet of metal or thick paper
Gamma	One Neutron	Neutral	Stopped by several inches of lead or lead of concrete

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No change	+1
Gamma	No change	No change

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



7. Complete the following nuclear equations:



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