



Name: \_\_\_\_\_  
Period: \_\_\_\_\_

KEY

## Nuclear Radiation Worksheet

1. a. What determines whether an atom is stable or unstable?

Its overall size and a difference between proton # and neutron #

- b. How does an unstable atom gain stability?

It undergoes radioactive decay

- c. What part of an atom undergoes change during radioactive decay?

The nucleus!

2. Which of the human senses can detect radiation?

Our eyes can detect colors, but we have no sense to detect alpha, beta, or gamma radiation

3. Explain the difference between an isotope and a radioactive isotope.

An isotope has a different # of protons / neutrons. Radioactive is large, unstable, and has a different # of p

4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

| Particle | Mass                  | Charge       | Penetrating power                                     |
|----------|-----------------------|--------------|---|
| Alpha    | 4 protons, 4 neutrons | Positive (+) | Stopped by paper or skin                              |
| Beta     | 0 No mass (electron)  | Negative (-) | Stopped by sheets of foil or thick magazine           |
| Gamma    | 0 No mass             | Neutral      | Stopped by several inches of lead or feet of concrete |

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

| Particle | Mass Number Change | Atomic Number Change |
|----------|--------------------|----------------------|
| Alpha    | -4                 | -2                   |
| Beta     | No change          | +1                   |
| Gamma    | No change          | No change            |

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



7. Complete the following nuclear equations:



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