

10.4 Electrophilic Reactions

Problem 10.41 is an electrophilic reaction, starting without strong base present. **Problem 10.42** is not.

Problem 10.41

Each of the following reactions is an electrophilic reaction. Draw the mechanism for each and label it.

- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .
- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .

Write the mechanism for each reaction and label it as electrophilic or nucleophilic.

- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .
- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .

When reacting H_2SO_4 with H_2O , the following reaction occurs:

Problem 10.42

- In the following mechanism, is H_2O acting as a nucleophile or electrophile? H_2O is acting as a nucleophile.
- Is H_2O acting as a nucleophile or electrophile? H_2O is acting as a nucleophile.

- In the following mechanism, is H_2O acting as a nucleophile or electrophile? H_2O is acting as a nucleophile.
- Is H_2O acting as a nucleophile or electrophile? H_2O is acting as a nucleophile.

- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .
- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .

Problem 10.43

- Reaction of water (nucleophile) with H_2SO_4 (electrophile) to form H_3O^+ and HSO_4^- .

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S Ashworth



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