

Stoichiometry

Solve the following problems:

1. Four moles of iron (II) sulfide added to seven moles of oxygen, produced two moles of iron (III) oxide and four moles of sulfur dioxide. (20pts)

a) Translate the problem to chemical equation.



Reactant side
4 Fe atoms
4 S atoms
 $7(2) = 14$ O atoms

Product side
4 Fe atoms
4 S atoms
 $2(3) + 4(2) = 14$ O atoms

b) How many grams of oxygen will react with 2.0 grams of iron (II) sulfide?

Given:

$m \text{ FeS} = 2.0 \text{ g}$
Req'd. = $m \text{ O}_2$

Solution:

molar mass of $\text{FeS} = 88 \text{ g/mol}$
mole ratio = 4 mol FeS : 7 mol O_2
molar mass of $\text{O}_2 = 32 \text{ g/mol}$

$$m_{\text{O}_2} = 2.0 \text{ g FeS} \left(\frac{1 \text{ mol FeS}}{88 \text{ g FeS}} \right) \left(\frac{7 \text{ mol O}_2}{4 \text{ mol FeS}} \right) \left(\frac{32 \text{ g O}_2}{1 \text{ mol O}_2} \right) = 1.3 \text{ g O}_2$$

2. Rusting involves the reaction of iron, Fe with oxygen, O_2 to form iron (III) oxide, Fe_2O_3 . (20pts)

a) Write the balanced chemical equation.



Reactant side
4 Fe atoms
 $3(2) = 6$ O atoms

Product side
4 Fe atoms
 $2(3) = 6$ O atoms

b) What are the mole ratios of the product to each reactant and of the reactants to each other?

Reactant to each other:

$$4 \text{ mol Fe} : 3 \text{ mol O}_2$$

Product to each reactant:

$$2 \text{ mol Fe}_2\text{O}_3 : 4 \text{ mol Fe}$$

Answers To Stoichiometry 1 With 3 Problems

Richard H. Eastman

Answers To Stoichiometry 1 With 3 Problems:

Relevant Problems for Chemical Principles Ian S. Butler,Arthur E. Grosser,1984 **Chemistry, Problem-Solving Worktext** James E. Brady, Frederick Senese, 2002-06-13 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical substances and the changes both chemical and physical that they undergo. Fireworks One of the ingredients is phosphorus a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms. The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light. Carbon The element is found in several forms including the familiar diamond and another recently discovered sooty substance that consists of soccer ball shaped molecules often referred to as buckeyballs. Diamond is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth. Acetylene torch Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of C_2H_2 and O_2 to give CO_2 and H_2O . Hot air balloon The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms. Stormy weather The evaporation of water serves to store energy provided by the sun. Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet. *Chemical Engineering Review Manual* Randall N.

Robinson,1980 **Big Data Analytics** Satish Narayana Srirama, Jerry Chun-Wei Lin, Raj Bhatnagar, Sonali Agarwal, P. Krishna Reddy, 2022-01-01 This book constitutes the proceedings of the 8th International Conference on Big Data Analytics BDA 2021 which took place during December 2021. Due to COVID 19 pandemic the conference was held virtually. The 16 full and 3 short papers included in this volume were carefully reviewed and selected from 41 submissions. The contributions were organized in topical sections named as follows medical and health applications machine deep learning IoTs sensors and networks fundamentation pattern mining and data analytics **Resources in Education** ,1997-05 **Problems and Solutions**

Guide to Accompany Rawn Biochemistry Robert Lindquist,1990 **Instructor's Manual for General Chemistry, Experiment and Theory** Richard H. Eastman,1970 **A Complete Preparation for the MCAT** Aftab S. Hassan,James L. Flowers,1992 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities. There are details on setting up a self managed study programme with guidelines for time management and stress management. All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress **Industrial Stoichiometry** Warren Kendall Lewis,Arthur Hitchcock Radasch,1926 Fuels and combustion Gas producers Sulfur compounds Metallurgy Crystallization **Water Treatment Plant Operation: Chemistry for operators** ,1981 **Problem Exercises for General Chemistry** G.Gilbert Long,Forrest C. Hentz,1986-08-14 A question answer supplement to accompany a 1 year general chemistry course. This text is a multiple choice format and can be used with any

standard general chemistry text The exercises emphasize the importance of problem solving and equation writing in the style used in general chemistry examinations and professional school aptitude examinations such as the MCAT and VCAT

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Complete Preparation for the MCAT Williams & Wilkins Review,1998-04 Here is the most respected test prep book for the Medical College Admission Test you can buy featuring an active learning approach for a better understanding of the exam's content and a better chance for success Unique to this guide are coverage of all recent changes in the MCAT plus a step by step plan for sharpening cognitive skills developing problem solving skills and critical thinking This thorough guide replaces expensive test preparation courses while giving students exactly what they need to get ready for the MCAT

Journal of the Ceramic Society of Japan ,1988 The Software Encyclopedia ,1988 New Technical Books New York Public Library,1969 **Basic Chemistry** G. William Daub,William S. Seese,1996 Written in a style and language that users without science backgrounds can understand This best selling introduction to the basic principles of chemistry draws on the reader's own experiences through analogies and cartoons to learn difficult concepts The clear systematic thinking approach to problem solving has also been highly praised by reviewers and users alike Countdown sections in each chapter consisting of five review questions keyed to previous material provide readers with a basis for material introduced in the new chapter Study exercises found immediately after new topics are introduced reinforce chapter problem material You and Chemistry marginal application icon relates chemistry to the real world End of chapter essays entitled Elements and Compounds relate the applications of specific elements or compounds to the readers life *Teaching Science for Understanding* James Joseph Gallagher,2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities **Resources in Education** ,1984 **Study Guide to Accompany Chemical Principles, Third Edition** Daniel L. Reger,1985

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