

25

NUCLEAR CHEMISTRY

SECTION 25.1 NUCLEAR RADIATION (pages 799–802)

This section describes the nature of radioactivity and the process of radioactive decay. It characterizes alpha, beta, and gamma radiation in terms of composition and penetrating power.

► Radioactivity (pages 799–800)

1. Which French chemist noticed that uranium salts could fog photographic plates, even without being exposed to sunlight?

2. What name did Marie Curie give to the process by which materials give off rays capable of fogging photographic plates? _____
3. An isotope that has an unstable nucleus is called a(n) _____.
4. Complete the table below to show basic differences between chemical and nuclear reactions.

Type of Reaction	Is Nucleus of Atom Changed?	Is Reaction Affected by Temperature, Pressure, or Catalysts?
Chemical		
Nuclear		

5. Complete the flowchart below, which describes the radioactive decay process.

The presence of too many or too few _____ relative to protons leads to an unstable nucleus.



At some point in time, an unstable nucleus will undergo a reaction and lose energy by emitting _____.



During the process of radioactive decay, an _____ radioisotope of one element is transformed eventually into a _____ isotope of a different element.

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