

**NUCLEAR CHEMISTRY****SECTION 25.1 NUCLEAR RADIATION (pages 799–802)**

This section describes the nature of radioactivity and the process of radioactive decay. It characterizes alpha, beta, and gamma radiation in terms of composition and penetrating power.

► **Radioactivity (pages 799–800)**

1. Which French chemist noticed that uranium salts could fog photographic plates, even without being exposed to sunlight? \_\_\_\_\_
2. What name did Marie Curie give to the process by which materials give off rays capable of fogging photographic plates? \_\_\_\_\_
3. An isotope that has an unstable nucleus is called a(n) \_\_\_\_\_.
4. Complete the table below to show basic differences between chemical and nuclear reactions.

Type of Reaction	Is Nucleus of Atom Changed?	Is Reaction Affected by Temperature, Pressure, or Catalysts?
Chemical	_____	_____
Nuclear	_____	_____

5. Complete the flowchart below, which describes the radioactive decay process.

The presence of too many or too few \_\_\_\_\_ relative to protons leads to an unstable nucleus.

At some point in time, an unstable nucleus will undergo a reaction and lose energy by emitting \_\_\_\_\_.

During the process of radioactive decay, an \_\_\_\_\_ radioisotope of one element is transformed eventually into a \_\_\_\_\_ isotope of a different element.

## Answers To Review Nuclear Chemistry Section 2

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