

Electronegativity and Polarity

Study Guide

Key Concepts

- The electronegativity difference determines the character of a bond between atoms.
- Polar bonds occur when electrons are not shared equally forming a dipole.
- The spatial arrangement of polar bonds in a molecule determines the overall polarity of a molecule.
- Molecules attract each other by weak intermolecular forces. In a covalent network solid, each atom is covalently bonded to many other atoms.

Answers To Electronegativity And Polarity Study Guide

Rachel Sandford

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