

Labflow Pre Lab 6 Questions

Which compounds could be represented by the empirical formula CH₂?

Select one or more:

- a) C₂H₆
- b) C₂H₄
- c) C₃H₆
- d) C₈H₁₈
- e) C₂H₂
- f) C₈H₁₆ - ✓✓b,c,f

Determine whether or not each chemical formula is an empirical formula.

- a) Ba₃(PO₄)₂
- b) K₂C₂O₄
- c) C₆H₁₂O₆
- d) CaCO₃
- e) NaMnO₄ - ✓✓a) Empirical
b) Not empirical
c) not empirical
d) Empirical
e) Empirical

To calculate the empirical formula, start by (1)

using (2). Then, (3) by (4). - ✓✓(1)converting the mass of each element to moles, (2) each elements molar mass, (3) calculate the ratio between the elements, (4)dividing the moles of each element by the smallest number of moles present.

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EW Minium

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