

Labflow Pre Lab 6 Questions

Which compounds could be represented by the empirical formula CH_2 ?

Select one or more:

- a) C_2H_6
- b) C_2H_4
- c) C_3H_6
- d) C_8H_{18}
- e) C_2H_2
- f) C_8H_{16} - ✓✓b,c,f

Determine whether or not each chemical formula is an empirical formula.

- a) $\text{Ba}_3(\text{PO}_4)_2$
- b) $\text{K}_2\text{C}_2\text{O}_4$
- c) $\text{C}_6\text{H}_{12}\text{O}_6$
- d) CaCO_3
- e) NaMnO_4 - ✓✓a) Empirical
- b) Not empirical
- c) not empirical
- d) Empirical
- e) Empirical

To calculate the empirical formula, start by (1)

using (2). Then, (3) by (4). - ✓✓(1)converting the mass of each element to moles, (2) each elements molar mass, (3) calculate the ratio between the elements, (4)dividing the moles of each element by the smallest number of moles present.

Answers To Chemthink Questions Precipitates Lab

EW Minium



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