

12. Use your answers to the CTQs above to calculate the value of K_c for $\text{Mg}(\text{OH})_2$.

13. Under what circumstances when Mg^{2+} (aq) and OH^- (aq) are present in a solution is the concentration of $\text{Mg}(\text{OH})_2$ (s) not equal to 40.5 mole/L?

Information

Notice that for the dissolution of any solid the expression for K_c will always contain the concentration of the solid in the denominator. Because the concentration of the solid is always constant (at a particular temperature), the concentration of the solid can be combined with K_c as follows (for $\text{Mg}(\text{OH})_2$):

Answers To Chemactivity 41

**Robert C. Smoot,Smoot,Richard G.
Smith,Jack Price**

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Organic Chemistry, a Guided Inquiry Andrei Straumanis,2004 **Chemistry** Richard S. Moog,John J. Farrell,2017-06-26
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