

12. Use your answers to the CTQs above to calculate the value of K_c for $\text{Mg}(\text{OH})_2$.
13. Under what circumstances when $\text{Mg}^{2+}(\text{aq})$ and $\text{OH}^{-}(\text{aq})$ are present in a solution is the concentration of $\text{Mg}(\text{OH})_2(\text{s})$ not equal to 40.5 mole/L?

Information

Notice that for the dissolution of any solid the expression for K_c will always contain the concentration of the solid in the denominator. Because the concentration of the solid is always constant (at a particular temperature), the concentration of the solid can be combined with K_c as follows (for $\text{Mg}(\text{OH})_2$):

Answers To Chemactivity 41

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



Answers To Chemactivity 41:

Organic Chemistry, a Guided Inquiry Andrei Straumanis, 2004 **Chemistry** Richard S. Moog, John J. Farrell, 2017-06-26

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