

Name: \_\_\_\_\_  
Date/Period: \_\_\_\_\_

### Atomic Number and Mass Number

Complete the following chart and answer the questions below.

| Element Name | Atomic Number | Number of Protons | Number of Neutrons | Mass Number |
|--------------|---------------|-------------------|--------------------|-------------|
| carbon       |               |                   |                    | 12          |
|              | 6             |                   | 6                  |             |
| hydrogen     |               |                   |                    | 1           |
|              |               | 1                 |                    | 4           |
| hydrogen     |               |                   | 2                  |             |
| nitrogen     |               |                   |                    | 14          |
|              |               |                   | 1                  | 2           |
|              | 92            |                   | 146                |             |
| Potassium    |               |                   | 20                 |             |
|              | 11            |                   | 12                 |             |
|              |               | 47                |                    | 108         |
| Xenon        |               |                   | 77                 |             |
|              |               |                   | 45                 | 80          |
|              |               | 24                |                    | 52          |
|              |               |                   | 16                 | 32          |
| Gold         |               |                   |                    | 197         |
|              | 76            |                   | 114                |             |

How are the atomic number and the number of protons related to each other?

How do the number of protons, number of neutrons, and the mass number relate to each other?

What is the one thing that determines the identity of an atom?

What sub atomic parts are equal to make a neutral atom?

Explain in your words how to calculate the mass of a sodium atom.

Draw and label three element symbols that are representative of the periodic table. It should have atomic #, mass #, name and element symbol.

# Atomic Number Activity Answer

**Philip L. Groisser**

## Atomic Number Activity Answer:

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