

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many water molecules form a hydrogen bond with a neighboring water molecule at typical temperatures? 25°C?
- ☐ 1
☒ 4
☐ 2
☐ 3

2. At equilibrium $H_2O(l) \rightleftharpoons H_2O(g)$, where would water molecules be located more frequently?
- ☒ liquid
☐ gas
☐ both
☐ none

3. The condensed state closest to the normal one:
- ☐ that of the elements
☐ that of the compounds
☒ that of the elements
☒ that of the compounds

4. Pure hexamethylsilane, $(CH_3)_6Si$, is a liquid at 25°C. How many water molecules are hydrogen-bonded to a single hexamethylsilane molecule at that temperature?
- ☐ 10
☐ 100
☐ 1000
☒ 10000

5. Water is the only substance that is more dense in its condensed state than in its gaseous state.
- ☐ because of hydrogen bonding
☒ because of hydrogen bonding
☐ because of hydrogen bonding
☐ because of hydrogen bonding

6. Water is the only liquid that is more dense in its condensed state than in its gaseous state.
- ☐ because of hydrogen bonding
☐ because of hydrogen bonding
☒ because of hydrogen bonding
☐ because of hydrogen bonding

7. A compound's condensed state is more dense in its condensed state than in its gaseous state. Which of the following is not a liquid at 25°C?
- ☒ H_2O
☒ H_2O
☐ H_2O
☐ H_2O

8. Pure hexamethylsilane, $(CH_3)_6Si$, is a liquid at 25°C. How many water molecules are hydrogen-bonded to a single hexamethylsilane molecule at that temperature?
- ☒ 10000
☐ 1000
☐ 100
☐ 10

9. At equilibrium $H_2O(l) \rightleftharpoons H_2O(g)$, where would water molecules be located more frequently?
- ☐ liquid
☐ gas
☒ liquid
☐ gas

10. Hexamethylsilane is a liquid at 25°C. How many water molecules are hydrogen-bonded to a single hexamethylsilane molecule at that temperature?
- ☐ 10000
☒ 10000
☐ 1000
☐ 100

11. The condensed state closest to the normal one:
- ☐ that of the elements
☐ that of the compounds
☒ that of the elements
☐ that of the compounds

Multiple Choice Stoichiometry Test

Leopold E. Klopfer



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