

## Atomic Structure & Isotope

### Part A: Atomic Structure

1. Complete the following table.

| Element name | Atomic number | Mass number | Number of protons | Number of electrons | Number of neutrons | Element symbol |
|--------------|---------------|-------------|-------------------|---------------------|--------------------|----------------|
|              |               | 70          |                   |                     |                    |                |
|              | 48            |             |                   |                     |                    |                |
|              |               |             |                   | 80                  |                    |                |
|              |               |             | 35                |                     |                    |                |

2. Identify the neutral atom described by name and mass number (example, oxygen-16).

i) The atom with 2 neutrons and 1 proton is \_\_\_\_\_.

ii) The atom with 17 electrons and 18 neutrons is \_\_\_\_\_.

iii) The atom with 6 protons and 8 neutrons is \_\_\_\_\_.

### Part B: Isotope

1. Complete the following table.

| Symbol            | # of protons | # of neutrons | # of electrons | Symbol                       | # of protons | # of neutrons | # of electrons |
|-------------------|--------------|---------------|----------------|------------------------------|--------------|---------------|----------------|
| ${}^1\text{H}_1$  |              |               |                | ${}^{17}\text{O}_8$          |              |               |                |
| ${}^2\text{H}_1$  |              |               |                | ${}^{39}\text{K}_{19}$       |              |               |                |
| ${}^4\text{He}_2$ |              |               |                | ${}^{40}\text{Ca}_{20}$      |              |               |                |
| ${}^6\text{He}_2$ |              |               |                | ${}^{238}\text{U}_{92}$      |              |               |                |
| ${}^6\text{Li}_3$ |              |               |                | ${}^{244}\text{Pu}_{94}$     |              |               |                |
| ${}^7\text{Li}_3$ |              |               |                | ${}^{56}\text{Fe}_{26}^{3+}$ |              |               |                |

# Atoms And Isotopes Worksheet Answers

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A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the American Geological Institute text.

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