

# Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

- a) Which element is the atom shown above? \_\_\_\_\_
- b) What is the chemical symbol of the element? \_\_\_\_\_
- c) What is the mass of the atom shown above? \_\_\_\_\_
- d) How many valence electrons are shown in the atom above? \_\_\_\_\_

② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? \_\_\_\_\_

③ The nucleus was discovered by which scientist:

- a) Thomson
- b) Dalton
- c) Rutherford
- d) Democritus

④ What is the mass of Carbon-14: \_\_\_\_\_

⑤ What is the number of protons in Carbon-14: \_\_\_\_\_

⑥ What is the number of neutrons in Carbon-14: \_\_\_\_\_

⑦ What is the number of protons in the radioactive Uranium-238: \_\_\_\_\_

⑧ What is the number of neutrons in Sulfur-33: \_\_\_\_\_

⑨ Neutral atoms of the same element can differ in

- a) Mass number
- b) Atomic number
- c) Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

- a) Avogadro
- b) Bohr
- c) Dalton



# Atomic Structure Chapter Worksheet

**Alyssa Fox-Charles**

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