

## Section 1 Foundations

PQ-1. The speed of light is  $3.0 \times 10^8 \text{ nm}\cdot\text{ns}^{-1}$ . What is this speed in  $\text{nm}\cdot\text{ms}^{-1}$ ?

(A)  $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$       (B)  $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$       (C)  $3.0 \times 10^{18} \text{ nm}\cdot\text{ms}^{-1}$       (D)  $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

(A) 1.2 g      (B) 1.20 g

Mass of beaker and water	29.62 g
Mass of beaker only	28.3220 g

(C) 1.298 g      (D) 1.2980 g

PQ-5. What is the name of  $\text{Ti}_2(\text{PO}_4)_3$ ?

(A) titanium phosphate  
 (B) titanium(III) phosphate  
 (C) titanium(IV) phosphate  
 (D) titanium tetrrophosphate

PQ-6. The formula of strontium hexafluorosilicate is  $\text{SrSiF}_6$ . What is the formula of aluminum hexafluorosilicate?

(A)  $\text{AlSiF}_6$       (B)  $\text{Al}_2(\text{SiF}_6)_3$       (C)  $\text{Al}_3\text{SiF}_6$       (D)  $\text{Al}_6(\text{SiF}_6)_2$

PQ-8. What is the correct name for  $\text{N}_2\text{O}_5$ ?

(A) nitrogen(II) oxide  
 (B) nitrogen(V) oxide  
 (C) dinitrogen oxide  
 (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of  $12.0 \text{ g}\cdot\text{cm}^{-3}$ , what is the mass of a nanoparticle with a volume of  $1.84 \times 10^{-21} \text{ cm}^3$ ?

(A)  $2.21 \times 10^{-23} \text{ g}$       (B)  $1.63 \times 10^{-23} \text{ g}$       (C)  $6.52 \times 10^{-21} \text{ g}$       (D)  $1.53 \times 10^{-22} \text{ g}$

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**Bruce W. Tuckman**

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1. Understanding the eBook 70 Question Chemistry Final Exam Answers
  - The Rise of Digital Reading 70 Question Chemistry Final Exam Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying 70 Question Chemistry Final Exam Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 70 Question Chemistry Final Exam Answers
  - User-Friendly Interface

4. Exploring eBook Recommendations from 70 Question Chemistry Final Exam Answers
  - Personalized Recommendations
  - 70 Question Chemistry Final Exam Answers User Reviews and Ratings
  - 70 Question Chemistry Final Exam Answers and Bestseller Lists
5. Accessing 70 Question Chemistry Final Exam Answers Free and Paid eBooks
  - 70 Question Chemistry Final Exam Answers Public Domain eBooks
  - 70 Question Chemistry Final Exam Answers eBook Subscription Services
  - 70 Question Chemistry Final Exam Answers Budget-Friendly Options
6. Navigating 70 Question Chemistry Final Exam Answers eBook Formats
  - ePUB, PDF, MOBI, and More
  - 70 Question Chemistry Final Exam Answers Compatibility with Devices
  - 70 Question Chemistry Final Exam Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 70 Question Chemistry Final Exam Answers
  - Highlighting and Note-Taking 70 Question Chemistry Final Exam Answers
  - Interactive Elements 70 Question Chemistry Final Exam Answers
8. Staying Engaged with 70 Question Chemistry Final Exam Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 70 Question Chemistry Final Exam Answers
9. Balancing eBooks and Physical Books 70 Question Chemistry Final Exam Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 70 Question Chemistry Final Exam Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 70 Question Chemistry Final Exam Answers
  - Setting Reading Goals 70 Question Chemistry Final Exam Answers
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of 70 Question Chemistry Final Exam Answers
  - Fact-Checking eBook Content of 70 Question Chemistry Final Exam Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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