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## AP CHEMISTRY FINAL EXAM REVIEW 2

1. Is energy absorbed or released in an endothermic reaction?: Absorbed.
2. What is the sign of delta H for an exothermic reaction?: Exothermic reactions release energy. The sign of delta H is negative.
3. When you touch a flask and it feels hot, is this an endothermic or exothermic process?: Exothermic; heat is being given off.
4. Define specific heat capacity.: The amount of heat required to raise 1g of a substance to 1 degrees Celsius.
5.  $q = mc(\Delta T)$  ; What does "q" stand for?: Heat (kJ)
6.  $q = mc(\Delta T)$  ; What does "c" stand for?: Specific heat capacity (J / g degrees C)
7. What is a hydrogen ion donor called?: An acid.
8. What is a hydrogen ion acceptor called?: A base.
9. Feels slippery; acid or base?: Base.
10. Tastes sour; acid or base?: Acid.
11. Tastes bitter; acid or base?: Base.
12.  $\text{pH} < 7$  ; acid, base, or neutral?: Acid
13.  $\text{pH} > 7$  ; acid, base, or neutral?: Base
14. Define neutralization.: When an acid reacts with a base to form water and a salt.  $\text{pH} = 7$ .
15. Soap; acid or base?: Base (tastes bitter)
16. Ammonia, acid or base?: Base (drain cleaners are bases).
17. Lemonade; acid or base?: Acid (tastes sour)
18. How many valence electrons are in an atom of chlorine?: 7
19. How many valence electrons does aluminum have?: 3
20. Define Reaction Rate: The decrease in concentration of reactants or the increase in concentration of products with time.
21. How does increasing temperature affect the rate of reaction?: Increasing temperature causes the kinetic energy to increase. Thus, molecules are moving faster and colliding more frequently. This causes the rate of reaction to increase.
22. If I decrease the concentration of reactant, what happens to the rate of reaction?: Decreasing concentration leads to less reactants available for collision. Therefore, reaction rate decreases.
23. How does surface area influence reaction rate?: By increasing the surface area of reactant molecules, we are increasing the space or point of contact that these molecules have available

# 2nd Semester Chemistry Final Exam Review Answers

**William H. Brown, Frederick A.  
Bettelheim, Jerry March**



## 2nd Semester Chemistry Final Exam Review Answers:

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web sep 22 2023 strabismus is a condition where there is misalignment of the visual axes of the eyes it may be latent or manifest and if manifest it may be constant or intermittent it is a common cause of diplopia and visual confusion seeing



different objects in the same place in adults it is an important

**strabismus crossed eyes** **aoa american optometric association** - Apr 29 2023

web crossed eyes or strabismus is a condition in which both eyes do not look at the same place at the same time it usually occurs due to poor eye muscle control or farsightedness strabismus usually develops in infants and young children most often by age 3