

## KEY

AP CHEMISTRY MULTIPLE CHOICE QUESTIONS  
Unit 8 – Introduction to Equilibrium

The equilibrium constant for the gas phase reaction above is 95 at 25°C. What is the value of the equilibrium constant for the following reaction at 25°C?



(A) 95

(B) 95

(C)  $95^2$ (D)  $\frac{1}{95}$ (E)  $\frac{1}{(95)^2}$ 

2. For which of the following reaction mixtures at equilibrium would reducing the volume of the container at constant temperature cause the equilibrium to shift toward the products? *increase "P"*



Shift towards fewer gas

4.



When 4.00 mol each of  $X(g)$  and  $Y(g)$  are placed in a 1.00 L vessel and allowed to react at constant temperature according to the equation above, 6.00 mol of  $Z(g)$  is produced. What is the value of the equilibrium constant,  $K_c$ ?

(A) 3

(B) 6

(C) 8

(D) 16

(E) 36



X

Y

Z

M

 $2X = 6M$ 

X

M

3M

X

M

# 2002 Ap Chemistry Multiple Choice Answer Key

**L Cohen**

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**Index Medicus**, 2003 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings **The Princess at the Keyboard** Amanda Stent, Philip Lewis, 2009-03-01 Are you a girl or young woman aged 13-18 If so this book is for you Amanda Stent and Philip Lewis have written a gentle friendly and comprehensive introduction to computer science Each chapter covers one area of computer science and includes examples of how the computer science works sidebars that contain historical notes or ideas for you to explore and biographies of women in computer science The last chapter covers questions that you might have about becoming a computer scientist We hope that after reading this book you will want to join us in studying this uniquely beautiful and practical subject **The Mantle and Core** R.W. Carlson, 2005-11-22 Though largely inaccessible the geochemistry of Earth's mantle and core can be examined through a wide variety of approaches Volume 2 focuses first on remote sensing using evidence from cosmochemical seismic petrologic and geochemical approaches Mantle composition is then examined in detail through descriptions of mantle samples brought to Earth's surface through tectonic volcanic and volatile outgassing processes The volume concludes with examination of processes that modify the composition

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