

10.4 Electrophilic Reactions

General Introduction An electrophile is an electron-seeking species, reacting with a lone pair or a double bond.

Electrophilic Substitution

Electrophilic Substitution (E₁ or E₂) is a reaction in which an electrophile attacks a molecule, replacing a leaving group.

- It occurs in saturated hydrocarbons (alkanes) and in aromatic compounds (arenes).
- The electrophile attacks the carbon atom of the alkane or arene, forming a carbocation intermediate.

Electrophilic Substitution is a reaction in which an electrophile attacks a molecule, replacing a leaving group.

- The electrophile attacks the carbon atom of the alkane or arene, forming a carbocation intermediate.
- The leaving group then departs, restoring the original structure of the molecule.

When reacting with an alkene, it follows the following mechanism:

Electrophilic Addition Mechanism

- In the first step, the electrophile attacks the double bond of the alkene, forming a carbocation intermediate.
- The carbocation intermediate then reacts with a nucleophile, forming a new bond.
- The final product is a saturated molecule with the electrophile and nucleophile added across the original double bond.

- The final product is a saturated molecule with the electrophile and nucleophile added across the original double bond.

Electrophilic Addition

- Electrophilic Addition is a reaction in which an electrophile attacks a molecule, adding across a double bond.

19 4 Neutralization Reactions

Ralph H. Petrucci, Robert K. Wismer



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