

10.4 Electrophilic Reactions

General Introduction An electrophile is an electron-seeking species, reacting with a lone pair or a double bond.

Electrophilic Reactions

Electrophilic reactions involving alkenes are addition reactions where both ends of the

- alkene are attacked by an electrophile (E⁺) to form a carbocation intermediate (an alkyl cation).
- The carbocation is then attacked by a nucleophile (Nu⁻) to form the final product, an alkylated alkane.

Electrophilic reactions involving alkenes are addition reactions where both ends of the

- alkene are attacked by an electrophile (E⁺) to form a carbocation intermediate (an alkyl cation).
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When reacting alkenes with H⁺, the following reactions occur:

Electrophilic Addition of H⁺

- In the following mechanism, the alkene reacts with H⁺ to form a carbocation intermediate. Hydrochloric acid (HCl) reacts with 2-methyl-2-butene to form 2-chloro-2-methylbutane. Write the mechanism for this reaction.
- Write down the mechanism for the reaction of 2-methyl-2-butene with H⁺ to form 2-methyl-2-butanol.
- Write down the mechanism for the reaction of 2-methyl-2-butene with H⁺ to form 2-methyl-2-butanol.
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- Write down the mechanism for the reaction of 2-methyl-2-butene with H⁺ to form 2-methyl-2-butanol.

Electrophilic Addition of Br₂

- Write down the mechanism for the reaction of 2-methyl-2-butene with Br₂ to form 2,3-dibromo-2-methylbutane.

19 4 Neutralization Reactions Section Review

Walter M. Gibson, Henry H. Teitelbaum



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