

# Acids and Bases

## Neutralization Reactions Lab



### Reading, Lab, and Practice Bundle (Grades 6-9)

#### Acids and Bases

**What is a Neutralization**


A neutralization reaction is when an acid reacts with a base to form a salt and water.

This type of chemical reaction is important in everyday life – for treating an upset stomach or for balancing soil pH in gardens.


**What Are Acids and Bases**

Acids are substances that taste sour and can have a low pH.

Examples: lemon juice.




#### Matching Vocabulary



Directions: Match each word on the left with the correct definition or example on the right. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces  $\text{OH}^-$  ions. (C)
- 2) The product of an acid and base reaction, often found in the form of  $\text{NaCl}$ . (E)
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is. (D)
- 4) A chemical that changes color to show if a substance is acidic or basic. (F)
- 5) A substance with a pH less than 7 that produces  $\text{H}^+$  ions. (B)
- 6) A reaction between an acid and a base to form water and salt. (A)

#### Matching Vocabulary



NAME: \_\_\_\_\_

DATE: \_\_\_\_\_

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- 4) A chemical that changes color to show if a substance is acidic or basic. (F)
- 5) A substance with a pH less than 7 that produces  $\text{H}^+$  ions. (B)
- 6) A reaction between an acid and a base to form water and salt. (A)

Acid


Base

pH scale

Neutralization

Salt

Indicator



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