

9-3 Review and Reinforcement

Classifying chemical Reactions

On the line on the left, write the letter of the type of chemical reaction represented by each equation below.

1. $A + B \rightarrow AB$	a. decomposition reaction
2. $A + BOC \rightarrow AX + B$	b. direct combination reaction
3. $AX + BY \rightarrow AY + BX$	c. double-replacement reaction
4. $AB \rightarrow A + B$	d. single-replacement reaction

Describe in words each of the following types of reactions.

a. direct combination reaction

b. decomposition reaction

c. single-replacement reaction

d. double-replacement reaction

Decide whether each of the following equations represents a direct combination (a), a decomposition (b), a single-replacement (c), or a double-replacement (d) reaction. Write your answer on the line.

9. $CO_2 \rightarrow C + O_2$
10. $NaCl + AgNO_3 \rightarrow NaNO_3 + AgCl$
11. $S_8 + 8Cl \rightarrow 8SCl_2$
12. $BaCl_2 + 2NaOH \rightarrow 2NaCl + Ba(OH)_2$
13. $Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$
14. $-CH_4 \rightarrow C + 2H_2$
15. $Pb(NO_3)_2 + Mg \rightarrow Pb + Mg(NO_3)_2$

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**Colin R. Martin, Vinood B. Patel, Victor
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