

Section Review

Objectives

- Describe how the amounts of reactants and products change in a chemical system at equilibrium.
- Identify three stresses that can change the equilibrium position of a chemical system.
- Explain what the value of K_{eq} indicates about the position of equilibrium.

Vocabulary

- reversible reaction
- chemical equilibrium
- equilibrium position
- Le Chatelier's principle
- equilibrium constant (K_{eq})

Key Equation

$$K_{\text{eq}} = \frac{[\text{C}]^c \times [\text{D}]^d}{[\text{A}]^a \times [\text{B}]^b}$$

When $a\text{A} + b\text{B} \rightleftharpoons c\text{C} + d\text{D}$

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short sentence, or number.

18 2 Reversible Reactions And Equilibrium Worksheet

Answers

**Jarvis, Gerry, Alberta Educational
Communications Corporation**

18 2 Reversible Reactions And Equilibrium Worksheet Answers:

Chemical Kinetics with Mathcad and Maple Viktor Korobov, Valery Ochkov, 2011-05-26 The authors explain at length the principles of chemical kinetics and approaches to computerized calculations in modern software suites mathcad and maple Mathematics is crucial in determining correlations in chemical processes and requires various numerical approaches Often significant issues with mathematical formalizations of chemical problems arise and many kinetic problems can't be solved without computers Numerous problems encountered in solving kinetics calculations with detailed descriptions of the numerical tools are given Special attention is given to electrochemical reactions which fills a gap in existing texts not covering this topic in detail The material demonstrates how these suites provide quick and precise behavior predictions for a system over time for postulated mechanisms Examples i.e. oscillating and non isothermal reactions help explain the use of mathcad more efficiently Also included are the results of authors own research toward effective computations [Current Index to Journals in Education](#), 1977 *Reversible Reactions and Dynamic Equilibrium : Utilization Guide* Jarvis, Gerry, Alberta Educational Communications Corporation, 1993

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