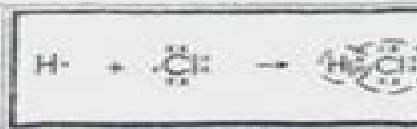


COVALENT BONDING

Name _____

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

H wants 2 electrons
all other atoms 8



Note that hydrogen is content with 2, not 8, electrons

6 electrons
in the middle
share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

1. H + H (H₂)
W = 2 - 4 - available
A = 2
S = 2



H - H
single bond - since 2 e⁻

2. F + F (F₂)
W = 8 + 8 = 16
A = 7 + 7 = 14
S = 2



:F-F:
single bond 2 e⁻

3. O + O (O₂)
W = 8 + 8 = 16
A = 6 + 6 = 12
S = 4



:O=O:
4 double bond

4. N + N (N₂)
W = 8 + 8 = 16
A = 5 + 5 = 10
S = 6



5. C + O (CO₂)
W = 8 + 8 = 24
A = 4 + 6 + 6 = 16
S = 8



6. H + O (H₂O)
W = 2 + 2 + 8 = 12
A = 1 + 1 + 6 = 8
S = 4



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James V. Quagliano

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1. Understanding the eBook 16 The Nature Of Covalent Bonding Answer Key
 - The Rise of Digital Reading 16 The Nature Of Covalent Bonding Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying 16 The Nature Of Covalent Bonding Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 16 The Nature Of Covalent Bonding Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from 16 The Nature Of Covalent Bonding Answer Key
 - Personalized Recommendations
 - 16 The Nature Of Covalent Bonding Answer Key User Reviews and Ratings

- 16 The Nature Of Covalent Bonding Answer Key and Bestseller Lists
- 5. Accessing 16 The Nature Of Covalent Bonding Answer Key Free and Paid eBooks
 - 16 The Nature Of Covalent Bonding Answer Key Public Domain eBooks
 - 16 The Nature Of Covalent Bonding Answer Key eBook Subscription Services
 - 16 The Nature Of Covalent Bonding Answer Key Budget-Friendly Options
- 6. Navigating 16 The Nature Of Covalent Bonding Answer Key eBook Formats
 - ePUB, PDF, MOBI, and More
 - 16 The Nature Of Covalent Bonding Answer Key Compatibility with Devices
 - 16 The Nature Of Covalent Bonding Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 16 The Nature Of Covalent Bonding Answer Key
 - Highlighting and Note-Taking 16 The Nature Of Covalent Bonding Answer Key
 - Interactive Elements 16 The Nature Of Covalent Bonding Answer Key
- 8. Staying Engaged with 16 The Nature Of Covalent Bonding Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 16 The Nature Of Covalent Bonding Answer Key
- 9. Balancing eBooks and Physical Books 16 The Nature Of Covalent Bonding Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 16 The Nature Of Covalent Bonding Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 16 The Nature Of Covalent Bonding Answer Key
 - Setting Reading Goals 16 The Nature Of Covalent Bonding Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 16 The Nature Of Covalent Bonding Answer Key
 - Fact-Checking eBook Content of 16 The Nature Of Covalent Bonding Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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