

## 16.4 Calculations Involving Colligative Properties

### Sample Problem 16.7

#### 1 Analyze List the knowns and the unknown.

- According to the definition of molality, the final solution must contain 0.060 mol KI per 1000 g  $\text{H}_2\text{O}$ .
- Use the molality as a conversion factor to convert from mass of the solvent ( $\text{H}_2\text{O}$ ) to moles of the solute (KI).
- Then use the molar mass of KI to convert from mol KI to g KI.

# 16 4 Calculations Involving Colligative Properties

## Answers

**Jicheng Xie**



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