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IGCSE A* MODEL ANSWER

0620 CHEMISTRY

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Paper 62



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1. (a) The student adds about 10 mL of dilute acid and followed by a few drops of aqueous silver nitrate.

answer ... **Ag⁺ + Cl⁻ → AgCl↓**

(b) To the second portion of solution C, the student adds about 10 mL of dilute acid and followed by a few drops of aqueous silver nitrate.

answer ... **Ag⁺ + Cl⁻ → AgCl↓**

(c) (i) To the third portion of solution C, the student adds a few drops of aqueous potassium hydroxide.

answer ... **Ag⁺ + OH⁻ → AgOH↓**

(ii) The student weighs the products from each of the three portions of
answer ... **AgOH** (white), **AgCl** (white), **Ag₂O** (black).

*	Red	White	Black
Initial Solution	Red	White	Black
Initial Paper (Red)	Red	White	Black
Initial Paper (White)	White	White	Black
Initial Orange	Red	Yellow	Orange
Residual solution	Colourless	Red	Colourless
Thymol solution	Colourless	Red	Colourless

Answer

Answer

Answer

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Lauren Gardner

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