

- 5 Carbonyl chloride, COCl_2 , is a colourless gas. It is made by the following reaction.



- (a) When the pressure on the equilibrium mixture is decreased, the position of equilibrium moves to left.
(b) How does the concentration of each of the three chemicals change?

_____ [2]

- (c) Explain why the position of equilibrium moves to left.

_____ [2]

- (d) Using the information given with the questions, is the forward reaction exothermic or endothermic? Give a reason for your choice.

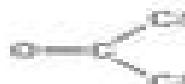
_____ [2]

- (e) Carbonyl chloride reacts with water to form two acidic compounds. Suggest which acidic compounds are formed.

1. _____ [2]

2. _____ [2]

- (f) The structural formula of carbonyl chloride is given below.



Draw a diagram that shows the arrangement of the valency electrons in one molecule of this covalent compound.

Use \times for an electron from a chlorine atom.

Use \circ for an electron from a carbon atom.

Use \oplus for an electron from an oxygen atom.

[Total: 12] [4]

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