



Cambridge Assessment
International Education

IGCSE A* MODEL ANSWER

0620 CHEMISTRY

Oct/Nov 2023

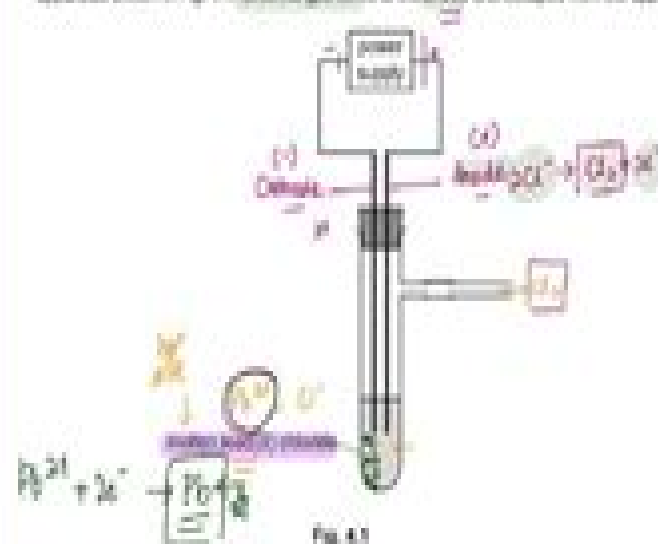
Paper 42



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(a) The student carries out an electrolysis experiment on molten lead(II) chloride using the apparatus shown in Fig. 4.1. (3) (Total gas taken at the anode and escapes from the apparatus.)



(i) Explain why lead(II) chloride needs to be molten before it will conduct electricity / react.
first moving electrons / mobile ions

(ii) Write the ionic half equation for the reaction occurring at the anode.



(iii) State the test for chlorine gas and describe the yellow fumes.

test Damp litmus paper

observations Bleached / turns white

(iv) Describe what is observed at the cathode.

0620 13 Oct Nov 2013 Chemistry Paper 3 Extended

DJ Losen



0620 13 Oct Nov 2013 Chemistry Paper 3 Extended:

0620 13 Oct Nov 2013 Chemistry Paper 3 Extended Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**0620 13 Oct Nov 2013 Chemistry Paper 3 Extended**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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