

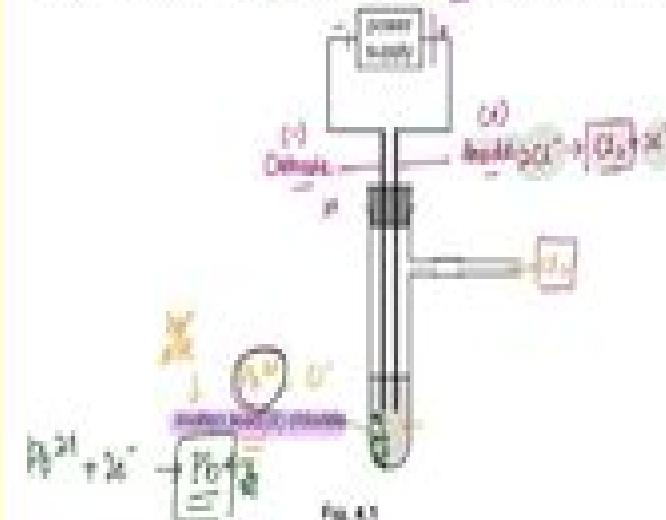


## IGCSE A\* MODEL ANSWER 0620 CHEMISTRY Oct/Nov 2023 Paper 42



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(a) The student carries out an electrolysis experiment on molten lead(II) chloride using the apparatus shown in Fig. 4.1. (Observe the labels at the ends and complete the equation.)



(b) Explain why lead(II) chloride needs to be molten before it will conduct electricity.  
first moving electrons / mobile ions

(c) Write the ionic half-equation for the reduction reaction of the anode.



(d) Give the name of the gas produced when the anode is immersed in water.

Chlorine / Cl2

Observation Clouded / Fumes white

(e) Describe what is observed at the cathode.

# 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended

**DJ Losen**

**0620 13 Oct Nov 2013 Chemistry Paper 3 Extended:**

## **0620 13 Oct Nov 2013 Chemistry Paper 3 Extended** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable.

This extraordinary book, aptly titled "**0620 13 Oct Nov 2013 Chemistry Paper 3 Extended**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Table of Contents 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended**

1. Understanding the eBook 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - The Rise of Digital Reading 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Advantages of eBooks Over Traditional Books
2. Identifying 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - User-Friendly Interface
4. Exploring eBook Recommendations from 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Personalized Recommendations
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended User Reviews and Ratings
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended and Bestseller Lists

5. Accessing 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended Free and Paid eBooks
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended Public Domain eBooks
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended eBook Subscription Services
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended Budget-Friendly Options
6. Navigating 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended eBook Formats
  - ePUB, PDF, MOBI, and More
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended Compatibility with Devices
  - 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Highlighting and Note-Taking 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Interactive Elements 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
8. Staying Engaged with 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
9. Balancing eBooks and Physical Books 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Setting Reading Goals 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Fact-Checking eBook Content of 0620 13 Oct Nov 2013 Chemistry Paper 3 Extended
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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