

Name :

Date :

# Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

i) Which element is the atom shown above? \_\_\_\_\_

ii) What is the chemical symbol of the element? \_\_\_\_\_

iii) What is the mass of the atom shown above? \_\_\_\_\_

iv) How many valence electrons are shown in the atom above? \_\_\_\_\_



② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? \_\_\_\_\_

③ The nucleus was discovered by which scientist:

a) Thomson

b) Dalton

c) Rutherford

d) Democritus

④ What is the mass of Carbon-14: \_\_\_\_\_

⑤ What is the number of protons in Carbon-14: \_\_\_\_\_

⑥ What is the number of neutrons in Carbon-14: \_\_\_\_\_

⑦ What is the number of protons in the radioactive Uranium-238: \_\_\_\_\_

⑧ What is the number of neutrons in Sulfur-33: \_\_\_\_\_

⑨ Neutral atoms of the same element can differ in

a) Mass number

b) Atomic number

c) Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

a) Avogadro

b) Bohr

c) Dalton

# Activity 147 Answers What Is An Atom

**David Baud**



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## **Table of Contents Activity 147 Answers What Is An Atom**

1. Understanding the eBook Activity 147 Answers What Is An Atom
  - The Rise of Digital Reading Activity 147 Answers What Is An Atom
  - Advantages of eBooks Over Traditional Books
2. Identifying Activity 147 Answers What Is An Atom
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Activity 147 Answers What Is An Atom
  - User-Friendly Interface
4. Exploring eBook Recommendations from Activity 147 Answers What Is An Atom

- Personalized Recommendations
  - Activity 147 Answers What Is An Atom User Reviews and Ratings
  - Activity 147 Answers What Is An Atom and Bestseller Lists
5. Accessing Activity 147 Answers What Is An Atom Free and Paid eBooks
    - Activity 147 Answers What Is An Atom Public Domain eBooks
    - Activity 147 Answers What Is An Atom eBook Subscription Services
    - Activity 147 Answers What Is An Atom Budget-Friendly Options
  6. Navigating Activity 147 Answers What Is An Atom eBook Formats
    - ePub, PDF, MOBI, and More
    - Activity 147 Answers What Is An Atom Compatibility with Devices
    - Activity 147 Answers What Is An Atom Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Activity 147 Answers What Is An Atom
    - Highlighting and Note-Taking Activity 147 Answers What Is An Atom
    - Interactive Elements Activity 147 Answers What Is An Atom
  8. Staying Engaged with Activity 147 Answers What Is An Atom
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Activity 147 Answers What Is An Atom
  9. Balancing eBooks and Physical Books Activity 147 Answers What Is An Atom
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Activity 147 Answers What Is An Atom
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Activity 147 Answers What Is An Atom
    - Setting Reading Goals Activity 147 Answers What Is An Atom
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Activity 147 Answers What Is An Atom

- 
- Fact-Checking eBook Content of Activity 147 Answers What Is An Atom
  - Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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