



# THE PROPERTIES OF MATTER

## Review Worksheet

1. What are the properties of matter? Properties of matter are the characteristics that distinguish one substance from another.

- a. What is mass? Mass is the amount of matter in an object. What is volume? Volume is the amount of space an object takes up. What is density? Density is the mass of an object divided by its volume. What is temperature? Temperature is a measure of the average kinetic energy of the particles in a substance. What is the SI unit for mass? Kilogram. What is the SI unit for volume? Cubic meter. What is the SI unit for density? Kilogram per cubic meter.
- b. What are the three states of matter? Solid, liquid, and gas. What are the properties of solids? Solids have a definite shape and volume. The particles in a solid are packed closely together and vibrate in place. Solids have a high density and are difficult to compress. What are the properties of liquids? Liquids have a definite volume but no definite shape. The particles in a liquid are close together but can move past each other. Liquids have a moderate density and are difficult to compress. What are the properties of gases? Gases have no definite shape or volume. The particles in a gas are far apart and move rapidly in all directions. Gases have a low density and are easy to compress.
- c. What is a mixture? A mixture is a combination of two or more substances that are not chemically combined. What are the properties of mixtures? Mixtures have the same chemical properties as the substances they are made of. The physical properties of a mixture are the sum of the physical properties of the substances it contains. What is a pure substance? A pure substance is a material that has a constant composition and chemical properties. What are the properties of pure substances? Pure substances have a constant composition and chemical properties. They can be elements or compounds.
- d. What is an element? An element is a substance that cannot be broken down into simpler substances by chemical means. What are the properties of elements? Elements have a constant composition and chemical properties. They are the building blocks of all matter. What is a compound? A compound is a substance that is made of two or more elements chemically combined in a fixed ratio. What are the properties of compounds? Compounds have a constant composition and chemical properties that are different from the elements they are made of. What is a mixture? A mixture is a combination of two or more substances that are not chemically combined. What are the properties of mixtures? Mixtures have the same chemical properties as the substances they are made of. The physical properties of a mixture are the sum of the physical properties of the substances it contains.
- e. What is a solid? A solid is a state of matter in which the particles are packed closely together and vibrate in place. Solids have a definite shape and volume. What are the properties of solids? Solids have a high density and are difficult to compress. They have a definite shape and volume. What is a liquid? A liquid is a state of matter in which the particles are close together but can move past each other. Liquids have a definite volume but no definite shape. What are the properties of liquids? Liquids have a moderate density and are difficult to compress. They have a definite volume but no definite shape. What is a gas? A gas is a state of matter in which the particles are far apart and move rapidly in all directions. Gases have no definite shape or volume. What are the properties of gases? Gases have a low density and are easy to compress. They have no definite shape or volume.

# 21 Properties Matter Answers

**Richard Bailey**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## 21 Properties Matter Answers:

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germain to the practice act and an appendix containing the acts concerning forcible entries T p      **The Codes of California as Amended and in Force at the Close of the Thirty-sixth Session of the Legislature, 1905 ...: Code of civil procedure** California, John F. MacLane, 1907      *A Digest of the Decisions Under the National Bankruptcy Act of 1898 Reported in the American Bankruptcy Reports, Volumes 1 to [27] Inclusive, (1898-1942) and of the Notes Therein Contained, with a Table of the Decisions Reported Therein and a Table of Sections of the Act of 1898, Construed In, Or Considered In, Or Affected By, Such Decisions* Melvin Thomas Bender, Harold Jay Hinman, 1909      *The popular educator* Popular educator, 1884

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