

LAB: SINGLE & DOUBLE REPLACEMENT REACTIONS



Single Replacement vs. Double Replacement Lab

Background: A single replacement reaction is one in which an element replaces an element in a compound. A double replacement reaction is one in which two elements swap with each other. Replace one another.

Materials:
Baking soda
Spoon

Popcorn
1 M Hydrochloric Acid

Combined Chlorine
0.5 M Potassium Iodide

Li	Be
Na	Mg
K	Ca
Rb	Sr
Fr	Ba

Procedure:

Part 1:
1. Add 5 mL of TACI to a test tube. Record observations.

2. Place a small piece of baking soda in the test tube. Record observations.

3. Write out a balanced chemical equation for both reactions.

Period

HW: Mixed Reactions Practice

Identify each chemical reaction as synthesis, decomposition, double replacement, or combustion reaction. Then balance the reaction.



Li	Be	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ge	As	Se	Br	Kr
Li	Be	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ge	As	Se	Br	Kr
Li	Be	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ge	As	Se	Br	Kr
Li	Be	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ge	As	Se	Br	Kr
Li	Be	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ge	As	Se	Br	Kr



10 2 Double Replacement Lab Answers

Michael Hoelscher

10 2 Double Replacement Lab Answers:

10 2 Double Replacement Lab Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**10 2 Double Replacement Lab Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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