

# Answer Key – Class 10 Science: Chemical Reactions and Equations

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## Section A: Multiple Choice Questions

1. c)  $Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$
2. b) Copper is deposited on iron
3. a) Hydrogen
4. d) All of the above
5. d)  $Zn + H_2SO_4 \rightarrow ZnSO_4 + H_2$

## Section B: Very Short Answer Questions

6.  $Ca + 2H_2O \rightarrow Ca(OH)_2 + H_2$
7. A precipitation reaction is a reaction where an insoluble solid (precipitate) is formed.  
Example:  
 $AgNO_3 + NaCl \rightarrow AgCl \downarrow + NaNO_3$
8. Colour changes from blue to light green. Equation:  $Fe + CuSO_4 \rightarrow FeSO_4 + Cu$
9. Double displacement reaction.  
Balanced:  $BaCl_2 + Na_2SO_4 \rightarrow BaSO_4 \downarrow + 2NaCl$
10. (i) Change in colour  
(ii) Evolution of gas / Formation of precipitate

## Section C: Short Answer Questions

11. A redox reaction involves both oxidation and reduction.  
Example:  $Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$   
Zn is oxidised (reducing agent).  $Cu^{2+}$  is reduced (oxidising agent).
12. On heating, lead nitrate decomposes to give lead oxide, nitrogen dioxide and oxygen.  
 $2Pb(NO_3)_2 \rightarrow 2PbO + 4NO_2 + O_2$   
Type: Thermal decomposition reaction.

## 10 1 Chemical Reactions Answer Key

**Thomas Edward Thorpe**

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## **Table of Contents 10 1 Chemical Reactions Answer Key**

1. Understanding the eBook 10 1 Chemical Reactions Answer Key
  - The Rise of Digital Reading 10 1 Chemical Reactions Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying 10 1 Chemical Reactions Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 10 1 Chemical Reactions Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from 10 1 Chemical Reactions Answer Key
  - Personalized Recommendations
  - 10 1 Chemical Reactions Answer Key User Reviews and Ratings
  - 10 1 Chemical Reactions Answer Key and Bestseller Lists
5. Accessing 10 1 Chemical Reactions Answer Key Free and Paid eBooks
  - 10 1 Chemical Reactions Answer Key Public Domain eBooks
  - 10 1 Chemical Reactions Answer Key eBook Subscription Services
  - 10 1 Chemical Reactions Answer Key Budget-Friendly Options

6. Navigating 10 1 Chemical Reactions Answer Key eBook Formats
  - ePUB, PDF, MOBI, and More
  - 10 1 Chemical Reactions Answer Key Compatibility with Devices
  - 10 1 Chemical Reactions Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 10 1 Chemical Reactions Answer Key
  - Highlighting and Note-Taking 10 1 Chemical Reactions Answer Key
  - Interactive Elements 10 1 Chemical Reactions Answer Key
8. Staying Engaged with 10 1 Chemical Reactions Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 10 1 Chemical Reactions Answer Key
9. Balancing eBooks and Physical Books 10 1 Chemical Reactions Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 10 1 Chemical Reactions Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 10 1 Chemical Reactions Answer Key
  - Setting Reading Goals 10 1 Chemical Reactions Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 10 1 Chemical Reactions Answer Key
  - Fact-Checking eBook Content of 10 1 Chemical Reactions Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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