

Activity Series Worksheet **KEY**

Use the activity series to determine if a reaction will occur. (A reaction will occur if the element **NOT** in the compound is higher on the activity series than the element that forms the same charged ion which is in the compound.) If a reaction will take place, predict the products and balance the reaction. If no reaction will occur, simply write, "no reaction". Notice the state of matter of your reactants and indicate the state of matter of your products. You may assume that all transition metals will form +2 ions.



Activity Series for Metals and Non-metals Single Replacement Reactions

Metal Name Symbol for ion	Decreasing activity
Lithium Li^+	reacts with water and acids
Potassium K^+	
Barium Ba^{2+}	
Strontium Sr^{2+}	
Calcium Ca^{2+}	
Sodium Na^+	
Magnesium Mg^{2+}	reacts with acids replacing hydrogen
Aluminum Al^{3+}	
Manganese Mn^{2+}	
Zinc Zn^{2+}	
Chromium Cr^{3+}	
Iron Fe^{2+}	
Cadmium Cd^{2+}	
Cobalt Co^{2+}	
Nickel Ni^{2+}	
Tin Sn^{2+}	
Lead Pb^{2+}	

Applying The Activity Series Answers

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Table of Contents Applying The Activity Series Answers

1. Understanding the eBook Applying The Activity Series Answers
 - The Rise of Digital Reading Applying The Activity Series Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Applying The Activity Series Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Applying The Activity Series Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Applying The Activity Series Answers
 - Personalized Recommendations
 - Applying The Activity Series Answers User Reviews and Ratings
 - Applying The Activity Series Answers and Bestseller Lists
5. Accessing Applying The Activity Series Answers Free and Paid eBooks
 - Applying The Activity Series Answers Public Domain eBooks
 - Applying The Activity Series Answers eBook Subscription Services
 - Applying The Activity Series Answers Budget-Friendly Options
6. Navigating Applying The Activity Series Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Applying The Activity Series Answers Compatibility with Devices
 - Applying The Activity Series Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Applying The Activity Series Answers
 - Highlighting and Note-Taking Applying The Activity Series Answers
 - Interactive Elements Applying The Activity Series Answers
8. Staying Engaged with Applying The Activity Series Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Applying The Activity Series Answers
- 9. Balancing eBooks and Physical Books Applying The Activity Series Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Applying The Activity Series Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Applying The Activity Series Answers
 - Setting Reading Goals Applying The Activity Series Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Applying The Activity Series Answers
 - Fact-Checking eBook Content of Applying The Activity Series Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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