

Oxidation Numbers

Periodic Table of the Elements

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1	1 H 1.0079	2 He 4.003																																											
2	3 Li 6.941	4 Be 9.01218																																											
3	5 B 10.811	6 C 12.011																																											
4	7 N 14.01	8 O 16.00																																											
5	9 F 18.998	10 Ne 20.1797																																											
6	11 Na 22.990	12 Mg 24.305	13 Al 26.982	14 Si 28.085	15 P 30.974	16 S 32.065	17 Cl 35.453	18 Ar 39.949																																					
7	19 K 39.098	20 Ca 40.078	21 Sc 44.956	22 Ti 47.867	23 V 50.942	24 Cr 52.00	25 Mn 54.938	26 Fe 55.845	27 Co 58.933	28 Ni 58.693	29 Cu 63.546	30 Zn 65.401	31 Ga 69.724	32 Ge 72.61	33 As 74.92	34 Se 78.96	35 Br 80.00	36 Kr 83.80																											
8	37 Rb 84.751	38 Sr 87.62	39 Y 88.902	40 Zr 91.224	41 Nb 92.906	42 Mo 95.94	43 Tc 97.907	44 Ru 98.907	45 Rh 98.907	46 Pd 98.907	47 Ag 107.868	48 Cd 112.411	49 In 114.84	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 I 126.90	54 Xe 131.30																											
9	55 Cs 132.91	56 Ba 137.34	57 La 138.906	58 Hf 140.912	59 Ta 145.00	60 W 145.75	61 Re 147.901	62 Os 151.901	63 Ir 152.901	64 Pt 157.847	65 Au 196.967	66 Hg 200.592	67 Tl 204.383	68 Pb 207.2	69 Bi 208.980	70 Po 210.0	71 At 214.0	72 Rn 222.0																											
10	73 Fr 223.0	74 Ra 226.0	75 Ac 227.0	76 Rf 228.0	77 Ha 229.0	78 106 251.0	79 107 252.0	80 108 253.0	81 109 254.0	82 110 255.0	83 110 256.0																																		
<ul style="list-style-type: none"> * Lanthanide Series * Actinide Series <table border="1"> <tbody> <tr> <td>52 Ce</td> <td>53 Pr</td> <td>54 Nd</td> <td>55 Pm</td> <td>56 Sm</td> <td>57 Eu</td> <td>58 Gd</td> <td>59 Tb</td> <td>60 Dy</td> <td>61 Ho</td> <td>62 Er</td> <td>63 Tm</td> <td>64 Yb</td> <td>65 Lu</td> </tr> <tr> <td>58 Th</td> <td>59 Pa</td> <td>60 U</td> <td>61 Np</td> <td>62 Pu</td> <td>63 Am</td> <td>64 Cm</td> <td>65 Bk</td> <td>66 Cf</td> <td>67 Es</td> <td>68 Fm</td> <td>69 Md</td> <td>70 No</td> <td>71 Lr</td> </tr> </tbody> </table>																		52 Ce	53 Pr	54 Nd	55 Pm	56 Sm	57 Eu	58 Gd	59 Tb	60 Dy	61 Ho	62 Er	63 Tm	64 Yb	65 Lu	58 Th	59 Pa	60 U	61 Np	62 Pu	63 Am	64 Cm	65 Bk	66 Cf	67 Es	68 Fm	69 Md	70 No	71 Lr
52 Ce	53 Pr	54 Nd	55 Pm	56 Sm	57 Eu	58 Gd	59 Tb	60 Dy	61 Ho	62 Er	63 Tm	64 Yb	65 Lu																																
58 Th	59 Pa	60 U	61 Np	62 Pu	63 Am	64 Cm	65 Bk	66 Cf	67 Es	68 Fm	69 Md	70 No	71 Lr																																

7 2 Oxidation Numbers Ws

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Sulfur: Compounds with sulfur of oxidation number IV

Ekkehard Fluck,1985 Bulletin of the Scientific and Technical Documentation Centre Markaz al-Qawmī lil-Buhūth (Egypt). Scientific and Technical Information Centre,1956 *Toxicology of the Nose and Upper Airways* John B.

Morris,Dennis J. Shusterman,2016-04-19 The application of molecular biologic methods recognition of neurogenic inflammatory processes and utilization of genetic knockout animals are just some of the advances in toxicology of the upper airways in recent years *Toxicology of the Nose and Upper Airways* presents a culmination of knowledge gained as a result of both human and experimental

Addison-Wesley Chemistry ,1990 Pandex Current Index to Scientific and Technical Literature ,1970 Official Gazette of the United States Patent and Trademark Office ,2000 Reagents for

Silicon-Mediated Organic Synthesis Philip L. Fuchs,2013-05-30 Over the last three decades the importance of organosiliconchemistry has greatly increased because it has opened a number ofnew synthetic strategies Silicon reagents are usually low cost versatile and allow a wide range of reactions This is the firstHandbook to compile essential Silicon containing reagents and makesuse of the leading reagent database e EROS Another hotvolume in the series Handbooks of Reagents for OrganicSynthesis this is a must have resource for all syntheticchemists working in drug development and

medicinal chemistry For the selection the Editor focussed on three key synthetic approaches with the greatest impact 1 Use of silicon as a temporary tether by unifying a reactive pair of functional groups and taking advantage of their template biased intramolecular cyclization 2 The specific use of the silane functionality as a hetero atom butyl group often colloquially referred to as the use of silicon as a fat proton 3 The use of the Brook rearrangement as an anion relay stratagem A new feature in this Handbook is the reagent finder analphabetically organized lookup table arranged by organic functionality and specific structure of the silicon atom to which it is bound Comprehensive Organometallic Chemistry Geoffrey Wilkinson, 1982

Engineering Record, Building Record and Sanitary Engineer, 1883 *Current Technical Papers*, 1974 *Atomic Industry Reporter: Laws and regulations* Bureau of National Affairs (Arlington, Va.), 1961 **Current List of Medical Literature**, 1941 **Proceedings of the National Academy of Sciences of the United States of America** National Academy of Sciences (U.S.), 2004 Indian Journal of Chemistry, 1999 A New English Dictionary on Historical Principles, 1888 **Journal of the Electrochemical Society** Electrochemical Society, 1960 Issues for 19 contain separately paged section consisting of editorials and various special features

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