

How to Factor Polynomials

Binomial

2 Terms

$$8x + 4$$

↓
GCF

$$4(2x + 1)$$

The factors of $8x + 4$ are:

$$4 \text{ & } (2x + 1)$$

Trinomial

3 Terms

$$a=1 \quad x^2 + 6x + 8$$

$$\boxed{(x+2)(x+4)}$$

$$a \neq 1 \quad 2x^2 - x - 6$$

$$\begin{array}{c|c} 2x^2 & -x - 6 \\ \hline & -3x - 6 \end{array}$$

$$2x(x+2) - 3(x+2)$$

$$\boxed{(2x-3)(x+2)}$$

Cubic

4 Terms

$$3y^3 + y^2 + 18y^2 + 6$$

$$\boxed{(3y^3 + y^2) + 18y^2 + 6}$$

$$\begin{array}{c|c} y(3y^2 + 1) & + 6(3y^2 + 1) \\ \hline & \end{array}$$

$$\boxed{(y+6)(3y^2 + 1)}$$

Algebra Factoring Polynomials Part Gcf Only Answers

Tom Carson, Ellyn Gillespie

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element click the card to flip atom click the card to flip 1 28

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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web mar 6 2017 1 a beryllium atom has 4 protons 5 neutrons and 4 electrons what is the mass number of this atom 4 5 8 9 13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5 neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the molecular structure are trigonal bipyramidal c both the electron geometry and the molecular structure are linear d both the electron geometry and the molecular structure are trigonal planar

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web a an oxygen atom with 8 protons and 8 neutrons b a potassium atom with 19 protons and 20 neutrons c a lithium atom with 3 protons and 4 neutron give complete symbols of each atom including the atomic number and the mass number

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