

How to Factor Polynomials

Binomial

2 Terms

$$8x + 4$$

↓ GCF

$$4(2x + 1)$$

The factors of $8x + 4$ are:

$$4 \text{ \& } (2x + 1)$$

Trinomial

3 Terms

$a=1$ $x^2 + 6x + 8$

Factors
 $(x+2)(x+4)$

$a \neq 1$ $2x^2 - x - 6$

$$2x^2 + 4x - 3x - 6$$

$$2x(x+2) - 3(x+2)$$

Factors
 $(2x-3)(x+2)$

Cubic

4 Terms

$$3y^3 + y + 18y^2 + 6$$

$$(3y^3 + y) + (18y^2 + 6)$$

$$y(3y^2 + 1) + 6(3y^2 + 1)$$

Factors
 $(y+6)(3y^2+1)$

Algebra Factoring Polynomials Part Gcf Only Answers

Tom Carson, Ellyn Gillespie



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Essential Algebra Carol Lee Johnston, Alden T. Willis, Jeanne Lazaris, 1988 This new ADVANTAGE SERIES of C L Johnston Alden T Willis and Jeanne Lazaris ESSENTIAL ALGEBRA is a traditional straight forward extremely popular book which is noted for its one step one concept at a time approach All major topics are divided into small sections each with its own examples and often with its own exercises an approach that helps students master each section before proceeding to the next one As part of the ADVANTAGE SERIES this version will offer all the quality content you ve come to expect from Johnston Willis and Hughes sold to your students at a significantly lower price

Elementary Intermediate Algebra: Student Solutions Manual Michael Sullivan, 2006-06-28

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web 1 draw five protons in the nucleus of the atom label them with their charge 2 draw six neutrons in the nucleus of the atom 3 draw two electrons in the first energy level and label them with their charge 4 draw three electrons in the second energy level and label them with their charge 5 what element is represented by the diagram

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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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web mar 6 2017 1 a beryllium atom has 4 protons 5 neutrons and 4 electrons what is the mass number of this atom 4 5 8 9 13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5 neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the molecular structure are trigonal bipyramid c both the electron geometry and the molecular structure are linear d both the electron geometry and the molecular structure are trigonal planar

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web a an oxygen atom with 8 protons and 8 neutrons b a potassium atom with 19 protons and 20 neutrons c a lithium atom with 3 protons and 4 neutron give complete symbols of each atom including the atomic number and the mass number

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