

Nuclear Chemistry (used in activities)

Model: Models represent the three isotopes of carbon:



Critical Thinking Questions

1. How many protons are in carbon-12? $\frac{6}{12}$ How many neutrons are in carbon-12? $\frac{6}{12}$
2. How many protons are in carbon-13? $\frac{6}{13}$ How many neutrons are in carbon-13? $\frac{7}{13}$
3. How many protons are in carbon-14? $\frac{6}{14}$ How many neutrons are in carbon-14? $\frac{8}{14}$
4. What is the smallest is the same property that is different among isotopes?
(Note: in physics, $q = \frac{m}{m_p}$)
 $\frac{1}{12}$ Atomic mass unit (amu) based
5. What specific subatomic composition is all three models based on (the Protons & Neutrons)?

Information: Nuclear reactions and binding reactions

A nuclear reaction is a change in the composition of the nucleus of an atom. That is not necessarily accompanied in chemical reaction, and that that depend on what substances are being mixed for.

There are three types of nuclear reactions: fusion, fission, and radioactive decay. Fusion is the joining of nuclei into larger nuclei, such as the stars where the sun and fusion. Fission the other way where a nucleus breaks up into smaller pieces as the process of nuclear fission.

Radioactive decay is spontaneous, meaning that there would happen decay of nuclei and give off particles, "rays," or heat. There is one thing that all particles which neutrons are radioactive.

Table 1: Some types of nuclear reactions producing nuclear reactions

Type of Reaction	Symbol	Mass Number	Charge	Relative Penetration Ability	Shielding Required	Biological Impact
Alpha particles	α , ${}^4_2\text{He}$	4	2+	very low	nothing	none unless inhaled
Beta particles	β , ${}^0_{-1}\text{e}$	0	1+	low	heavy cloth, plastic	slightly to quite high
Gamma rays	γ , ${}^0_0\gamma$	0	0	very high	lead or concrete	extremely high
Neutrons	n , ${}^1_0\text{n}$	1	0	very high	water, lead	extremely high
Protons	p , ${}^1_1\text{H}$	1	1+	low	heavy cloth, plastic	slightly to quite high

Answers Chemactivity 6

**Michael P. Garoutte, Ashley B.
Mahoney**



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Introductory Chemistry Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting General, Organic, and Biological Chemistry Michael P. Garoutte, 2014-02-24

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Prentice Hall Chemistry, 2000

Chemistry Richard S. Moog, John J. Farrell, 2008-01-09 The ChemActivities found in Chemistry A Guided Inquiry Fourth Edition use the classroom guided inquiry approach and provide an excellent accompaniment to Spencer's Chemistry Structure and Dynamics Fourth Edition or any other General Chemistry text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting They are designed to train students to use and analyze data figures and text to deduce chemical concepts **Process Oriented Guided Inquiry Learning**

(POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL's theoretical basis its implementations in diverse environments and evaluation of student outcomes Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry

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