

Student Exploration: Solubility and Temperature

Vocabulary:

1. concentration - Molar concentration is a measure of the concentration of a chemical species, in particular of a solute in a solution, in terms of amount of substance per unit volume of solution.
2. dissolve - dissolve is to cause a solute to pass into a solution.
3. homogeneous mixture - A homogeneous mixture is a solid, liquid, or gaseous mixture that has the same proportions of its components throughout any given sample.
4. solubility - Solubility is the property of a solid, liquid or gaseous chemical substance called solute to dissolve in a solid, liquid or gaseous solvent.
5. solubility curve - A solubility curve is a data based graph comparing the amount of solute that will dissolve in a given amount of solvent at various temperatures.
6. solute - the minor component in a solution, dissolved in the solvent.
7. solution - a liquid mixture in which the minor component (the solute) is uniformly distributed within the major component (the solvent).
8. solvent - able to dissolve other substances.

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. What happens when you stir a spoonful of sugar into hot water?
The sugar dissolves into the water.
2. When sugar or another substance is **dissolved** in water, it disappears from view and forms a **homogeneous mixture** with the water, also called a **solution**. The water tastes sweet and may appear slightly cloudy.
If you can't see the sugar, how can you tell that it is there?
3. Does sugar dissolve more easily in hot water or cold water? **Hot water**

Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the *Solubility and Temperature* Gizmo, you will study how temperature affects how much solute will dissolve in a solution.

To begin, check that **Potassium nitrate** is selected and the **Temp.** of the water is 20 °C. Click **OK**.



1. In this solution, what is the solute? **Potassium nitrate** What is the solvent? **Water**
2. Click **Add 10 g** to mix 10 g of potassium nitrate into the water.
 - A. Did all of the potassium nitrate dissolve? **yes**
 - B. How can you tell? **All of the potassium nitrate disappeared as it was stirred into the water.**

Answers For Student Exploration Solubility And Temperature

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In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Answers For Student Exploration Solubility And Temperature** a literary masterpiece penned with a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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Table of Contents Answers For Student Exploration Solubility And Temperature

1. Understanding the eBook Answers For Student Exploration Solubility And Temperature
 - The Rise of Digital Reading Answers For Student Exploration Solubility And Temperature
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Student Exploration Solubility And Temperature
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Student Exploration Solubility And Temperature
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Student Exploration Solubility And Temperature
 - Personalized Recommendations

- Answers For Student Exploration Solubility And Temperature User Reviews and Ratings
- Answers For Student Exploration Solubility And Temperature and Bestseller Lists
- 5. Accessing Answers For Student Exploration Solubility And Temperature Free and Paid eBooks
 - Answers For Student Exploration Solubility And Temperature Public Domain eBooks
 - Answers For Student Exploration Solubility And Temperature eBook Subscription Services
 - Answers For Student Exploration Solubility And Temperature Budget-Friendly Options
- 6. Navigating Answers For Student Exploration Solubility And Temperature eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Student Exploration Solubility And Temperature Compatibility with Devices
 - Answers For Student Exploration Solubility And Temperature Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Student Exploration Solubility And Temperature
 - Highlighting and Note-Taking Answers For Student Exploration Solubility And Temperature
 - Interactive Elements Answers For Student Exploration Solubility And Temperature
- 8. Staying Engaged with Answers For Student Exploration Solubility And Temperature
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Student Exploration Solubility And Temperature
- 9. Balancing eBooks and Physical Books Answers For Student Exploration Solubility And Temperature
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Student Exploration Solubility And Temperature
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Student Exploration Solubility And Temperature
 - Setting Reading Goals Answers For Student Exploration Solubility And Temperature
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Student Exploration Solubility And Temperature
 - Fact-Checking eBook Content of Answers For Student Exploration Solubility And Temperature

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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