

Medical Laboratory Science Review - Harr – Chemistry |81 questions with correct answers

Which formula correctly describes the relationship between absorbance and %T ?

- A. $A = 2 - \log \%T$
- B. $A = \log 1/T$
- C. $A = -\log T$
- D. All of these options Answer ✓ ✓ D Absorbance is proportional to the inverse log of transmittance.

$$A = -\log T = \log 1/T$$

Multiplying the numerator and denominator by 100 gives:

$$A = \log (100/100 \times T)$$

$100 \times T = \%T$, substituting %T

for $100 \times T$ gives:

$$A = \log 100/\%T$$

$$A = \log 100 - \log \%T$$

$$A = 2.0 - \log \%T$$

For example, if %T = 10.0, then:

$$A = 2.0 - \log 10.0$$

$$\log 10.0 = 1.0$$

$$A = 2.0 - 1.0 = 1.0$$

A solution that has a transmittance of 1.0 %T

Answers For Lab 6b Heath Chemistry

B Lingard

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