

Note:

- i. All question are compulsory
- ii. Answer to the two sections are to be written in the same answer book.
- iii. Figure to the right hand side indicate full marks.
- iv. Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- v. Every new question must be started on a new page.
- vi. Use of logarithmic table is allowed

CHEMISTRY: SECTION – I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- i. In body centred cubic structure the space occupied is about
(A) 68 % (B) 53 %
(C) 38 % (D) 32 %
- ii. For a gaseous reaction the unit of rate of reaction is
(A) L atm s^{-1} (B) $\text{atm mol}^{-1} \text{ s}^{-1}$
(C) atm s^{-1} (D) mol s
- iii. Which of the following compounds contains $\text{S}=\text{O}$ as well as $\text{S}=\text{S}$ bonds?
(A) Sulphuric acid (B) Thiosulphuric acid
(C) Sulphurous acid (D) Thiosulphurous acid
- iv. Which of the following solutions shows maximum depression in freezing point?
(A) 0.5 M Li_2SO_4 (B) 1 M NaCl
(C) 0.5 M $\text{Al}_2(\text{SO}_4)_3$ (D) 0.5 M BaCl_2
- v. For a chemical reaction $\Delta S = -0.035 \text{ kJ/K}$ and $\Delta H = -204 \text{ kJ}$.
(A) 5.14 K (B) 57.14 K
(C) 571.4 K (D) 5714.0 K
- vi. The standard e.m.f of the following cell is 0.463 V
 $\text{Cu} \mid \text{Cu}^{2+}_{(\text{aq})} \parallel \text{Ag}^{+}_{(\text{aq})} \mid \text{Ag}$. If $E_{\text{Ag}}^{\circ} = 0.800 \text{ V}$,
What is the standard potential of Cu electrode?
(A) 1.137 V (B) 0.337 V
(C) 0.463 V (D) -0.463 V
- vii. Fe_2O_3 is reduced to spongy iron near the top of blast furance by
(A) H_2 (B) CaO
(C) SiO_2 (D) CO

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- Chapter 3: Answers For Exercise Of Hsc 11th Chemistry in Everyday Life
- Chapter 4: Answers For Exercise Of Hsc 11th Chemistry in Specific Contexts
- Chapter 5: Conclusion

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Table of Contents Answers For Exercise Of Hsc 11th Chemistry

1. Understanding the eBook Answers For Exercise Of Hsc 11th Chemistry
 - The Rise of Digital Reading Answers For Exercise Of Hsc 11th Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Exercise Of Hsc 11th Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Exercise Of Hsc 11th Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Exercise Of Hsc 11th Chemistry
 - Personalized Recommendations
 - Answers For Exercise Of Hsc 11th Chemistry User Reviews and Ratings
 - Answers For Exercise Of Hsc 11th Chemistry and Bestseller Lists
5. Accessing Answers For Exercise Of Hsc 11th Chemistry Free and Paid eBooks
 - Answers For Exercise Of Hsc 11th Chemistry Public Domain eBooks
 - Answers For Exercise Of Hsc 11th Chemistry eBook Subscription Services
 - Answers For Exercise Of Hsc 11th Chemistry Budget-Friendly Options
6. Navigating Answers For Exercise Of Hsc 11th Chemistry eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answers For Exercise Of Hsc 11th Chemistry Compatibility with Devices
 - Answers For Exercise Of Hsc 11th Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Exercise Of Hsc 11th Chemistry
 - Highlighting and Note-Taking Answers For Exercise Of Hsc 11th Chemistry
 - Interactive Elements Answers For Exercise Of Hsc 11th Chemistry

8. Staying Engaged with Answers For Exercise Of Hsc 11th Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Exercise Of Hsc 11th Chemistry
9. Balancing eBooks and Physical Books Answers For Exercise Of Hsc 11th Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Exercise Of Hsc 11th Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Exercise Of Hsc 11th Chemistry
 - Setting Reading Goals Answers For Exercise Of Hsc 11th Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Exercise Of Hsc 11th Chemistry
 - Fact-Checking eBook Content of Answers For Exercise Of Hsc 11th Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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 - Integration of Multimedia Elements
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