

Chemistry Benchmark Assessment

Student Name: _____

1. Which of the following reactions represents a synthesis reaction?

- ☐ A. $2 \text{H}_2\text{O}(\text{g}) \rightarrow 2 \text{H}_2(\text{g}) + \text{O}_2(\text{g})$
- ☐ B. $\text{CaCO}_3(\text{s}) + 2 \text{HNO}_3(\text{aq}) \rightarrow \text{Ca(NO}_3)_2(\text{aq}) + 2 \text{H}_2\text{O}(\text{l})$
- ☐ C. $\text{Zn}(\text{s}) + \text{CuSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Cu}(\text{s})$
- ☐ D. $2 \text{Mg}(\text{s}) + \text{O}_2(\text{g}) \rightarrow 2 \text{MgO}(\text{s})$

2. When a nitrogen atom covalently bonds to three hydrogen atoms to form a molecule, it will retain one pair of unshared electrons. According to VSEPR theory, what is the shape of an ammonia molecule?

- ☐ A. trigonal planar
- ☐ B. linear tetrahedral
- ☐ C. pyramidal
- ☐ D. tetrahedral

3. An unknown isotope has 17 neutrons and an atomic mass of 32. What is the name of this element?

- ☐ A. phosphorus
- ☐ B. sodium
- ☐ C. sulfur
- ☐ D. chlorine

4. Which of the following does not possess the same number of valence electrons?

- ☐ A. elements within the same period
- ☐ B. all elements with the same atomic number
- ☐ C. all elements with the same atomic mass
- ☐ D. elements within the same group

Answers For Chemistry Benchmark

Jerri Lynn Amos Osmar



Answers For Chemistry Benchmark:

Terrestrial Microcosms and Environmental Chemistry James M. Witt, James W. Gillett, 1977 **Innovative Methods of Teaching and Learning Chemistry in Higher Education** Ingo Eilks, Bill Byers, 2015-11-06 Two recent initiatives from the EU namely the Bologna Process and the Lisbon Agenda are likely to have a major influence on European Higher Education. It seems unlikely that traditional teaching approaches which supported the elitist system of the past will promote the mobility, widened participation and culture of life long learning that will provide the foundations for a future knowledge based economy. There is therefore a clear need to seek new approaches to support the changes which will inevitably occur. The European Chemistry Thematic Network ECTN is a network of some 160 university chemistry departments from throughout the EU as well as a number of National Chemical Societies including the RSC which provides a discussion forum for all aspects of higher education in chemistry. This handbook is a result of one of their working groups who identified and collated good practice with respect to innovative methods in Higher Level Chemistry Education. It provides a comprehensive overview of innovations in university chemistry teaching from a broad European perspective. The generation of this book through a European Network with major national chemical societies and a large number of chemistry departments as members make the book unique. The wide variety of scholars who have contributed to the book make it interesting and invaluable reading for both new and experienced chemistry lecturers throughout the EU and beyond. The book is aimed at chemistry education at universities and other higher level institutions and at all academic staff and anyone interested in the teaching of chemistry at the tertiary level. Although newly appointed teaching staff are a clear target for the book, the innovative aspects of the topics covered are likely to prove interesting to all committed chemistry lecturers.

Environmental Toxicology and Chemistry, 2005 **Models and Measurements Intercomparison II**, 1999 Models and measurements intercomparison II MM II summarizes the intercomparison results from model simulations and observations of stratospheric species. Representatives from 23 modeling groups using 29 models participated in these MM II exercises between 1996 and 1999. Twelve of the models were two dimensional zonal mean models while 17 were three dimensional models. This was an international effort as seven were from outside the U S A. Six transport experiments and five chemistry experiments were designed for various models. Models participating in the transport experiments performed simulations of chemically inert tracers providing diagnostics for transport. The chemistry experiments involved simulating the distributions of chemically active trace gases including ozone.

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