

Name : _____

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STOICHIOMETRY PRACTICE WORKSHEET

1. Using the following equation:



Calculate how many grams of iodine are needed to prepare 28.6 grams of ICl by this reaction.

2. Using the following equation:



How many moles and grams of KMnO₄ are needed for this reaction on 11.4 grams of KNO₂?

3. Using the following equation:



How many moles and grams of oxygen (O₂) are needed to react with 56.8 grams of ammonia by this reaction?

4. Using the following equation:



Calculate the number of moles and the number of grams of iodine (I₂) that can be made this way from 16.4 grams of NaIO₃.

9 Stoichiometry Practice Problems Review Answers

Claire Aldridge, Karl Lee, Kaplan, Inc

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CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level

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