

PEPERIKSAAN PERCUBAAN SPM 2014
4541/3 CHEMISTRY
Paper 3

Question	Rubric	Score								
1 (a)	Able to record all the temperature accurately with one decimal place. <u>Answer</u>	3								
	<table><tr><th>Experiment I</th><th>Experiment II</th></tr><tr><td>Initial temperature = 29.0</td><td>Initial temperature = 29.0</td></tr><tr><td>Highest temperature = 41.0</td><td>Highest temperature = 39.0</td></tr><tr><td>Change of temperature = 12.0</td><td>Change of temperature = 10.0</td></tr></table>		Experiment I	Experiment II	Initial temperature = 29.0	Initial temperature = 29.0	Highest temperature = 41.0	Highest temperature = 39.0	Change of temperature = 12.0	Change of temperature = 10.0
	Experiment I		Experiment II							
	Initial temperature = 29.0		Initial temperature = 29.0							
	Highest temperature = 41.0	Highest temperature = 39.0								
Change of temperature = 12.0	Change of temperature = 10.0									
Able to record any 3 readings accurately // all readings correctly but without decimal point.	2									
Able to record any 2 readings correctly	1									
No response given / wrong response	0									

Question	Rubric	Score
1(b)	Able to state the relationship between temperature and type of alkali <u>Sample answer:</u> The temperature change is higher when sodium hydroxide solution react with hydrochloric acid // [vice versa]	3
	Able to state the relationship between temperature and type of alkali less correctly <u>Sample answer:</u> The temperature change is higher when strong alkali react with hydrochloric acid// Temperature is directly proportional with the strength of alkali	2
	Able to give an idea of the relationship between temperature and rate of reaction <u>Sample answer:</u> Type of alkali affects the temperature change	1
	No response or wrong response	0

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