

**PEPERIKSAAN PERCUBAAN SPM 2014**  
**4541/3 CHEMISTRY**  
**Paper 3**

Question	Rubric		Score							
	Able to record all the temperature accurately with one decimal place.									
	<u>Answer:</u>									
	<table border="1" style="display: inline-table; vertical-align: middle;"> <tr> <td><b>Experiment I</b></td> <td><b>Experiment II</b></td> </tr> <tr> <td>Initial temperature = 29.0</td> <td>Initial temperature = 29.0</td> </tr> <tr> <td>Highest temperature = 41.0</td> <td>Highest temperature = 39.0</td> </tr> <tr> <td>Change of temperature = 12.0</td> <td>Change of temperature = 10.0</td> </tr> </table>	<b>Experiment I</b>	<b>Experiment II</b>	Initial temperature = 29.0	Initial temperature = 29.0	Highest temperature = 41.0	Highest temperature = 39.0	Change of temperature = 12.0	Change of temperature = 10.0	3
<b>Experiment I</b>	<b>Experiment II</b>									
Initial temperature = 29.0	Initial temperature = 29.0									
Highest temperature = 41.0	Highest temperature = 39.0									
Change of temperature = 12.0	Change of temperature = 10.0									
1 (a)	Able to record any 3 readings accurately // all readings correctly but without decimal point.		2							
	Able to record any 2 readings correctly		1							
	No response given / wrong response		0							

Question	Rubric	Score
	Able to state the relationship between temperature and type of alkali	
	<u>Sample answer:</u> The temperature change is higher when sodium hydroxide solution react with hydrochloric acid // [vice versa]	3
1(b)	Able to state the relationship between temperature and type of alkali less correctly	
	<u>Sample answer:</u> The temperature change is higher when strong alkali react with hydrochloric acid// Temperature is directly proportional with the strength of alkali	2
	Able to give an idea of the relationship between temperature and rate of reaction	
	<u>Sample answer:</u> Type of alkali affects the temperature change	1
	No response or wrong response	0

## A18pp2 P3 Mark Scheme 2014

**J Spring**

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