

# Buffers Keep the Balance

Data Table 1. Model Carbonate Blood Buffer

Table 1	Test Tube Number					
	1	2	3	4	5	6
Solution	Carbonic Acid (Reference)	Model Carbonate Blood Buffer	Water (Control)	Model Carbonate Blood Buffer	Water (Control)	Sodium Bicarbonate (Reference)
Bromothymol blue indicator color	yellow <del>6.0</del>	greenish 7.2 blue	green <del>6.0</del>	greenish 7.0 blue	green <del>6.0</del>	blue
Initial pH value	6.0	7.2	6.0	7.0	6.0	8.0
Number of drops of HCl required to convert solutions 2 and 3 to yellow (acid reference color)	NA	16	5	NA	NA	NA
Number of drops of NaOH required to convert solutions 4 and 5 to blue (base reference color)	NA	NA	NA	2	2	NA

Data Table 2. Effect of HCl on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color			
		Initial	After 1 drop of HCl	After 5 drops of HCl	After 10 drops of HCl
Water (control)	6.0	6.0	6.0	NA	NA
Phosphate Buffer A	6.0	6.0	6.0	6.0	6.0
Phosphate Buffer B	6.4	6.4	6.4	6.0	6.0

Data Table 3. Effect of NaOH on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color			
		Initial	After 1 drop of NaOH	After 5 drops of NaOH	After 10 drops of HCl
Water (control)	6.0	6.0	7.0	NA	NA
Phosphate Buffer A	6.0	6.0	6.8	7.0	8.0
Phosphate Buffer B	6.4	6.4	7.0	7.0	7.0

# Answers Buffers Keep The Balance Lab Answers

**RC Schank**



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