

Chemistry Reference Sheet

Cations

Cation name	Formula
Sodium	Na^+
Potassium	K^+
Magnesium	Mg^{2+}
Calcium	Ca^{2+}
Iron (II)	Fe^{2+}
Iron (III)	Fe^{3+}

Anions

Anion name	Formula
Fluoride	F^-
Chloride	Cl^-
Bromide	Br^-
Iodide	I^-
Sulfide	S^{2-}
Selenide	Se^{2-}
Phosphide	P^{3-}
Carbide	C^{4-}

Covalent prefixes

Number of atoms of an element	Prefix
1*	mono-
2	di-
3	tri-
4	tetra-
5	penta-
6	hexa-
7	hepta-
8	octa-

*the prefix for one atom is only used for the second element

Cation name	Formula
Hydrogen	H^+
Copper (I)	Cu^+
Copper (II)	Cu^{2+}
Zinc (II)	Zn^{2+}
Aluminium	Al^{3+}
Ammonium	NH_4^+

Anion name	Formula
Hydride	H^-
Cyanide	CN^-
Acetate	CH_3COO^-
Bromate	BrO_3^-
Selenate	SeO_4^{2-}
Sulfate	SO_4^{2-}
Carbonate	CO_3^{2-}
phosphate	PO_4^{3-}

Common Compounds

Anion name	Formula
Ammonia	NH_3
Carbon dioxide	CO_2
Ethanol	$\text{C}_2\text{H}_5\text{OH}$
Hydrochloric acid	HCl
Hydrogen peroxide	H_2O_2
Nitric acid	HNO_3
Glucose	$\text{C}_6\text{H}_{12}\text{O}_6$
Sodium hydroxide	NaOH
Sulfuric acid	H_2SO_4
Water	H_2O

General Equations for Calculations

Density $d = \frac{m}{V}$ Heat energy $Q = mc\Delta T$

Error: $\text{error} = \text{measured} - \text{theoretical}$

%Error: $\text{\%error} = \frac{\text{measured} - \text{theoretical}}{\text{theoretical}} \times 100$

Act Quality Core Chemistry Reference Sheet

L Darling-Hammond



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