

Chemistry Reference Sheet

Common Cations

Cation name	Symbol
Sodium	Na^+
Proton	H^+
Magnesium	Mg^{2+}
Calcium	Ca^{2+}
Ammonium	NH_4^+
Boron (BH ₄ ⁻)	BH_4^-

Cation name	Symbol
Hydrogen	H^+
Chloride (Cl ⁻)	Cl^-
Chlorate (ClO ₃ ⁻)	ClO_3^-
Chromate (CrO ₄ ²⁻)	CrO_4^{2-}
Phosphate (PO ₄ ³⁻)	PO_4^{3-}
Perchlorate (ClO ₄ ⁻)	ClO_4^-

Common Anions

Anion name	Symbol
Fluoride	F^-
Chloride	Cl^-
Bromide	Br^-
Chromate	CrO_4^{2-}
Chromite	CrO_4^{2-}
Perchlorate	ClO_4^-
Chlorite	ClO_4^-

Anion name	Symbol
Hydroxide	OH^-
Carboxylate	CO_3^{2-}
Acetate	CH_3COO^-
Bisulfite	HSO_3^-
Sulfide	S^{2-}
Sulfate	SO_4^{2-}
Chlorate	ClO_3^-
Chlorite	ClO_4^-

Common Polyatomic Ions

Polyatomic ion name	Symbol
Hydroxide	OH^-
Carboxylate	CO_3^{2-}
Acetate	CH_3COO^-
Bisulfite	HSO_3^-
Sulfide	S^{2-}
Sulfate	SO_4^{2-}
Chlorate	ClO_3^-

Common Element Symbols

Element name	Symbol
Hydrogen	H
Carbon dioxide	CO_2
Chlorine	Cl
Hydrochloric acid	HCl
Hydrogen peroxide	H_2O_2
Sulfuric acid	H_2SO_4
Chromite	CrO_4^{2-}
Chromate	CrO_4^{2-}
Water	H_2O

*This page is for general use. It is only used for one assessment.

General Instructions for Calculations

$$\text{Density} = \text{m} = \frac{\text{m}}{\text{v}}$$

$$\text{Heat energy} = Q = m \cdot c \cdot \Delta T$$

Given: m = mass in grams and c = calorimetric constant

Unknown: ΔT = temperature change in degrees Celsius
 Q = heat energy in Joules

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