

CHEM 101 ACS Study Guide

Chapter 1

SQ-1. Planck's constant is 6.626×10^{-34} J·s. What is this value in J·μs?

(A) 6.626×10^{-17} J·μs (B) 6.626×10^{-19} J·μs
 (C) 6.626×10^{-14} J·μs (D) 6.626×10^{-16} J·μs

SQ-4. Twenty irregular pieces of an unknown metal have a collective mass of 28.225 g. When carefully placed in a graduated cylinder that originally contained 7.75 mL of water, the final volume read 11.70 mL. What is the density of the metal (in g·cm⁻³)?

(A) 2.41 g·cm⁻³ (B) 3.64 g·cm⁻³ (C) 7.15 g·cm⁻³ (D) 11.2 g·cm⁻³

PQ-12. If a palladium nanoparticle has a density of 12.0 g·cm⁻³, what is the mass of a nanoparticle with a volume of 1.54×10^{-21} cm³?

(A) 2.21×10^{-20} g (B) 1.63×10^{-20} g (C) 6.52×10^{-21} g (D) 1.53×10^{-22} g

PQ-13. Which sample has the largest volume?

Densities	
aluminum: 2.7 g·cm ⁻³	copper: 9.0 g·cm ⁻³
(A) 1.0 g aluminum	(B) 5.0 g aluminum
(C) 1.0 g copper	(D) 5.0 g copper

Chapter 2

SQ-3. Which compound is named correctly?

(A) FeCO₃, iron carbonate (B) K₂SO₄, potassium sulfate
 (C) Sr(NO₃)₂, strontium nitrite (D) Co₂(SO₄)₃, cobalt(II) sulfate

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