

CHEM 101 ACS Study Guide

Chapter 1

SQ-1. Planck's constant is 6.626×10^{-34} J-s. What is this value in kJ- μ s?

- (A) 6.626×10^{-43} kJ- μ s (B) 6.626×10^{-27} kJ- μ s
(C) 6.626×10^{-34} kJ- μ s (D) 6.626×10^{-21} kJ- μ s

SQ-8. Twenty irregular pieces of an unknown metal have a collective mass of 28.525 g. When carefully placed in a graduated cylinder that originally contained 7.75 mL of water, the final volume read 11.70 mL. What is the density of the metal (in g-cm³)?

- (A) 2.41 g-cm⁻³ (B) 3.64 g-cm⁻³ (C) 7.15 g-cm⁻³ (D) 11.2 g-cm⁻³

PQ-12. If a palladium nanoparticle has a density of 12.0 g-cm⁻³, what is the mass of a nanoparticle with a volume of 1.84×10^{-21} cm³?

- (A) 2.21×10^{-26} g (B) 1.63×10^{-26} g (C) 6.52×10^{-21} g (D) 1.53×10^{-22} g

PQ-13. Which sample has the largest volume?

Densities			
aluminum	2.7 g-cm ⁻³	copper	9.0 g-cm ⁻³

- (A) 1.0 g aluminum (B) 5.0 g aluminum (C) 1.0 g copper (D) 5.0 g copper

Chapter 2

SQ-3. Which compound is named correctly?

- (A) FeCO₃, iron carbonate (B) K₂SO₃, potassium sulfite
(C) Sn(NO₂)₂, stannous nitrite (D) Co₂(SO₄)₃, cobalt(II) sulfate

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